**TU KL Biophysics** Schünemann Lab



# Nuclear resonant scattering of chemical and biological systems with focussed beams and high resolution monochromators





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EBS-Workshop on Nuclear Resonance Scattering ESRF - Grenoble - France 11 & 12 March 2019



## 1. Iron(II) Spin-Crossover (SCO-) Complexes as Molecular Switches



•Molecules can be switched thermally or optically between the diamagnetic low spin (LS) S=0 and the paramagnetic high spin (HS) S=2 state

•Potential Application: Molecular Spintronics



Juliusz A. Wolny



Exploration of low energy phonon modes in chemical systems

Exploration of anharmonic effects which couple low energy vibrations with iron ligand modes

## energy phonon materials





distance: 5 mm with T=4.2 K at sample



## **Optical pump - nuclear resonance probe experiments on SCO complexes**



S. Sakshath, K. Jenni, L. Scherthan, P. Würtz, M. Herlitschke, I. Sergeev, C. Strohm, H.-C.Wille, R. Röhlsberger, J. A. Wolny, V. S. Hyperfine Interact (2017) 238: 89

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## Interaction of [Fe(PM-BiA)<sub>2</sub>(NCS)<sub>2</sub>] with laser pulses



### **Nuclear Forward Scattering (80K)**



Optical pump - nuclear resonance probe experiments on spin crossover complexes, S. Sakshath, K. Jenni, L. Scherthan, P. Würtz, M. Herlitschke, I. Sergeev, C. Strohm, H.-C. Wille, R. Röhlsberger, J. A. Wolny, V. S. Hyperfine Interact. (2017) 238: 89.

**Time dependent** pDOS in High-**Repetiton Optical Pump-NIS Probe** experiments (300 ps time resolution at present)

## clear resonance probe n SCO complexes



Optical pump – NIS probe (Δt=1 ns)







Exploration of spin phonon interaction in single molecule magnets

Search for molecular modes relevant to under barrier spin relaxation in single molecule magnets

## relaxation in Iron magnets by <sup>57</sup>Fe NIS







## **3. Dynamic Properties of chemical models for iron sites in proteins**



The Ferric Low-spin Heme Model [TPPFe<sup>III</sup>(NH<sub>2</sub>Pz)<sub>2</sub>]Cl



S=1/2

Schünemann V, Raitsimring AM, Benda R, Trautwein AX, Shokireva TK, Walker FA (1999) ESEEM and Mössbauer studies of the ferriheme model compound bis(3-amino-pyrazole)tetraphenylporphyrinato iron(III) chloride [TPP Fe(NH<sub>2</sub>PzH)<sub>2</sub>]Cl. JBIC 4, 708-716.

#### [TPPFe<sup>III</sup>(NH<sub>2</sub>Pz)<sub>2</sub>]Cl: Nuclear Resonance Vibrational Spectroscopy (NRVS) in DMF Solution and Influence of Solvent on Normal Modes





## [TPPFe<sup>III</sup>(NH<sub>2</sub>Pz)<sub>2</sub>]Cl: Nuclear Resonance Vibrational Spectroscopy (NRVS) in DMF Solution and Influence of Solvent on Normal Modes



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#### leme sliding motions

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Exploration of solvent effects in iron containing homogeneous catalysts and biological models

# cm<sup>-1</sup>

Solvent modes couple to heme sliding and doming motions

## **Exploration of anharmonic** effects

H. Auerbach et al. To be published

7.6 cm

IIOM field uff

## 4. The NO-transporter nitrophorin (NP): Binding of small signal molecules



B. Moeser, A. Janoschka, J.A. Wolny, H. Paulsen, I. Filippov, R.E. Berry, H. Zhang, A.I. Chumakov, F.A. Walker, V. S., J. Am. Chem. Soc. (2012) 134, 4216 TECHNISCHE UNIVERSITÄT KAISERSLAUTERN Exploration of the influence of the protein matrix on functional relevant iron ligand modes

# Also here: Exploration of anharmonic effects



## 5. LytB (IspH), a 4Fe-4S-Protein of the MEP Pathway Essential for Isoprenoid Biosynthesis in Pathogenic Bacteria





A. Ahrens-Botzong, K. Janthawornpong, J.A. Wolny, E.N. Tombou, M. Rohmer, S. Krasutzsky, C.D. Poulter, V. S., M. Seemann, *Biosynthesis of isoprene units. Mössbauer spectroscopy proofs on substrate- and inhibitor-binding to the [4Fe-4S] cluster of the LytB/IspH enzyme*, Angew. Chem. Int. Ed. (2011) 12182.

#### Simulation of NIS Data Leads to a Structural Model of the 4Fe-4S center of Substrate Free LytB







I. Faus et al. Angew. Chem. Int. Ed. (2015) 54, 12584





## Inhibitor-Enzyme X-Ray Structure



Spectroscopic proof (or disproof) of X-ray crystal structure data



6. Protonation of dimetallic cluster of the R2-like ligand-binding oxidase(R2lox) from *Geobacillus kaustophilus* 

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Stockholms universitet

J. J. Griese

V. Srinivas M. Högbom

R2lox exists as Mn<sup>III</sup>Fe<sup>III</sup>-R2lox and as Fe<sup>III</sup>Fe<sup>III</sup>R2lox!

- 1. How do metal proteins recognize their metals?
- 2. What is the protonation of these the dimetallic clusters?



## NIS of the R2-like ligand-binding oxidase from Geobacillus kaustophilus



Mn<sup>III</sup>Fe<sup>III</sup>-R2lox

0.010



**Jennifer Marx** 



R. Kositzki, S. Mebs, J. Marx, J. J. Griese, N. Schuth, M. Högbom, V. S., and M. Haumann, Inorg. Chem. (2016), 55 (19) 9869

## Exploration of low energy functional relevant iron ligand modes in proteins

## Exploration of the solvent: water

#### of the R2-like ligandacillus kaustophilus





J. Marx et al. to be published

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#### 7. NIS on Protein Single Crystals

Orientation Dependent NIS of a Myoglobin Compound II Protein Single Crystal





A new sample environment for cryogenic nuclear resonance scattering experiments on single crystals and microsamples at P01, PETRA III; Rackwitz, S., Faus, I., Schmitz, M. et al. Hyperfine Interact (2014) 226: 673.

## Phonons in chemical nanostructures (e.g. SCO materials)

## How does a surface influence molecular modes in SMMs

## ations: CO compounds





S. Rackwitz, I. Faus, B. Lägel, J. Linden, J. Marx, E. Oesterschulze, K. Schlage, H.-C. Wille, S. Wolff, J. A. Wolny, V. S., Hyperfine Interact. 226 (2014) 667-671.



## Search for active sites in **Complex Systems:**

## **Particles, Clusters, Single** iron ions

## cations: TECHNISCHE UNIVERSITÄT KAISERSLAUTERN eneous catalysts http://www.3dchem.com/i magesofmolecules/H-ZSM-5.jpg $NO_x$ to $N_2$ . B=5 T Fe<sup>III</sup>: S=5/2 (E/D=0<sup>\*</sup>) Fe<sup>III</sup>: S=5/2 (E/D=0.33<sup>\*</sup>) Fe<sup>III</sup>: S=2 Fe<sup>III</sup>-O-Fe<sup>III</sup>: S=0 **Superparamagnetic Nanoparticles**

10 0 5 pcity (mms⁻¹)

Cooperation with W. Grünert A. Brückner

\*E/D from EPR

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lez, I. Ellmers, H. Huang, U. Bentrup, V. S., inert, A. Brückner, J. Catal. 316, 103 (2014)

## Iron agglomeration in tissue

## cations prage in single



le





Fig. 4 Mössbauer spectra obtained at 4.2K (a) and 77K (b). Dashed lines represent contamination due to the minor, yet detectable, <sup>57</sup>Fe content of the windows in the cryostat. Solid lines were simulated on the basis of Lorentzians with the Mössbauer parameters as summarized in Table 1.

## Spatial resolved Iron transport in single cells

## plications storage in algae Ils





n algae *Ectocarpus siliculosus* cells d with a stain indicating presence of

Carl J. Carrano Department of Chemistry and Biochemistry San Diego State University

C. Schmidt Universität zu Lübeck

## **10. Future: Iron inside single cells**



Lindahl and coworkers: Biophysical Investigation of the Ironome of Human Jurkat Cells and Mitochondria Biochemistry, 2012, 51 (26), 5276.



Iron is essential for cell survival, but toxic if not properly regulated. This diagram shows the complexity of human iron homeostasis already in its simplified form. Iron plays a decicive role in cell metabolism, Cell Death, and Disease

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