

Introduction to autoclave technics

*In-situ studies of high-temperature fluids and melts ($P < 2$ kbar)
and their application to Geosciences*

@ BM16, BM30B and Institut Néel

Marion Louvel

Hands on ! High-pressure techniques at the ESRF-EBS – June 21st, 2019

Introduction : High-temperature fluids and associated challenges

Why studying high-temperature fluids?

High T fluids ($100 < T < 800 \text{ }^\circ\text{C}$) play a key role in geological processes and have important impact on our societies

- ⇒ *Volcanic degassing (Explosive vs. Passive ; Climatic impact)*
- ⇒ *Ore deposit formation*
- ⇒ *Petroleum reservoirs*
- ⇒ *Geothermal energy and CO₂ storage*



Pinatubo 1991 eruption, Indonesia



*Kawah-Ijen, Indonesia (S deposit)
Bingham, USA (Cu-Au-Mo porphyry)*



*Hellisheidi geothermal power plant
Carbfix CO₂ storage project, Iceland*

Why studying high-temperature fluids?

High T fluids ($100 < T < 800 \text{ }^\circ\text{C}$) play a key role in geological processes and have important impact on our societies

- ⇒ *Volcanic degassing (Explosive vs. Passive ; Climatic impact)*
- ⇒ *Ore deposit formation*
- ⇒ *Petroleum reservoirs*
- ⇒ *Geothermal energy and CO₂ storage*



Pinatubo 1991 eruption, Indonesia



Other (non-geological) applications:

Supercritical solvents for synthesis, extraction, *etc...*
Vapor-liquid, liquid-liquid equilibria

What we want to know

Composition

depends on nature of protolith and P-T

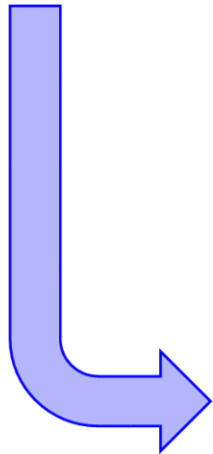
- Solubility of volatile-rich minerals in fluids
- Nature of volatiles species as a function of P-T (HCO₃⁻ or CO₃²⁻?)
- Phase diagram for complex fluids involving H₂O-CO₂-NaCl-F, S, P, etc...

What we want to know

Composition

depends on nature of protolith and P-T

- Solubility of volatile-rich minerals in fluids
- Nature of volatiles species as a function of P-T (HCO_3^- or CO_3^{2-} ?)
- Phase diagram for complex fluids involving H_2O - CO_2 - NaCl - F , S , P , etc...



Effect of fluid-rock-magma interactions on composition

- Solubility of metals and minerals
- Nature of metal complexation (e.g. Cu(II)Cl^{2-} or Cu(I)Cl_2^-)

Circulation in rocks/magmas

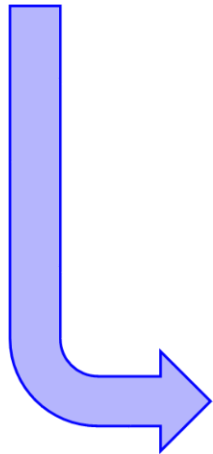
- Physical properties (density, viscosity)

What we want to know

Composition

depends on nature of protolith and P-T

- Solubility of volatile-rich minerals in fluids
- Nature of volatiles species as a function of P-T (HCO_3^- or CO_3^{2-} ?)
- Phase diagram for complex fluids involving H_2O - CO_2 - NaCl - F , S , P , etc...

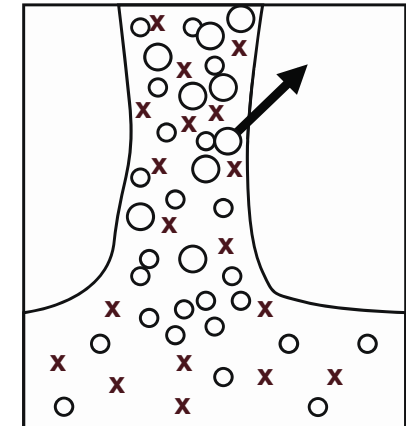


Effect of fluid-rock-magma interactions on composition

- Solubility of metals and minerals
- Nature of metal complexation (e.g. Cu(II)Cl_2^- or Cu(I)Cl_2^-)

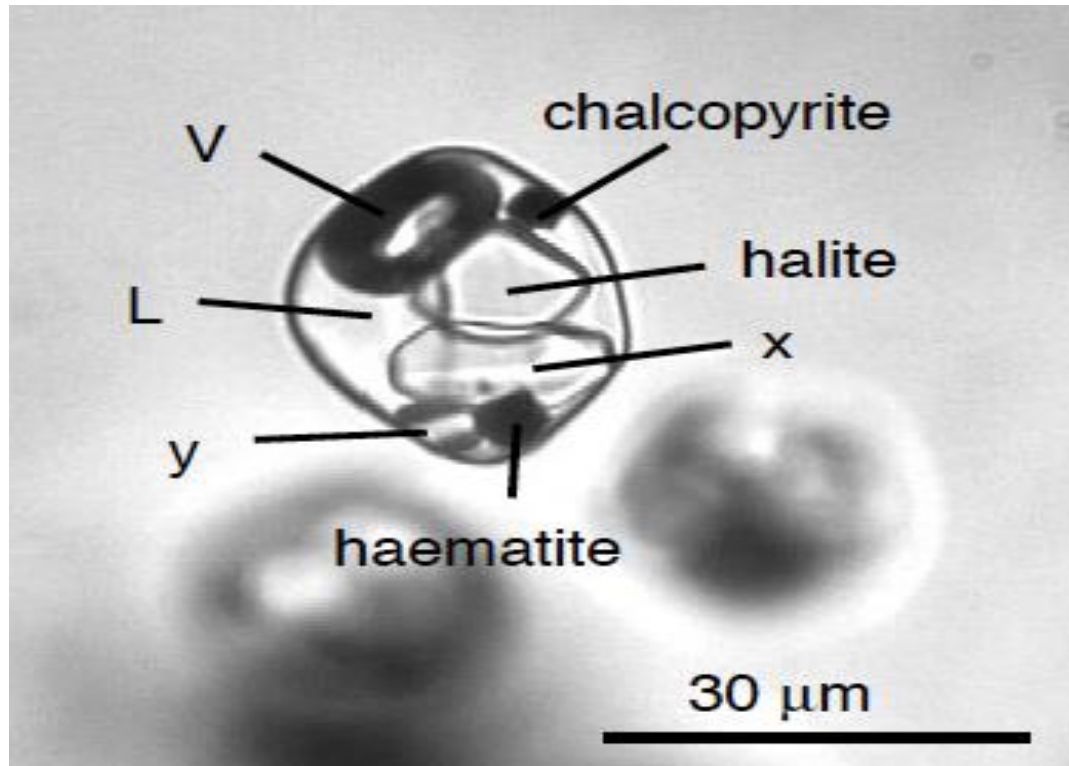
Circulation in rocks/magmas

- Physical properties (density, viscosity)



Challenging studies....

Fluids are difficult/impossible to sample....



Fluid inclusions

✗ **Post-entrapment modification**

Fast H₂O loss (eg. Bakker and Doppler, 2016)

Cu, Au diffusion (Lerchbaumer and Audetat, 2012; Guo and Audetat, 2017)



Volcanic gases

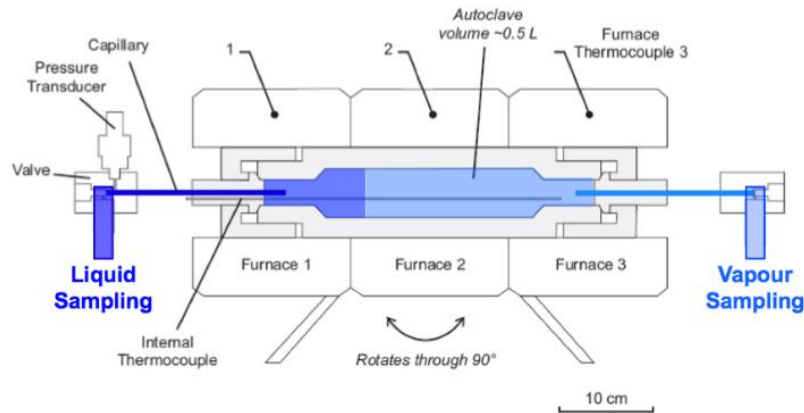
✗ **Do not represent 'deep' composition**

Challenging studies....

=> Experimental constraints are required !

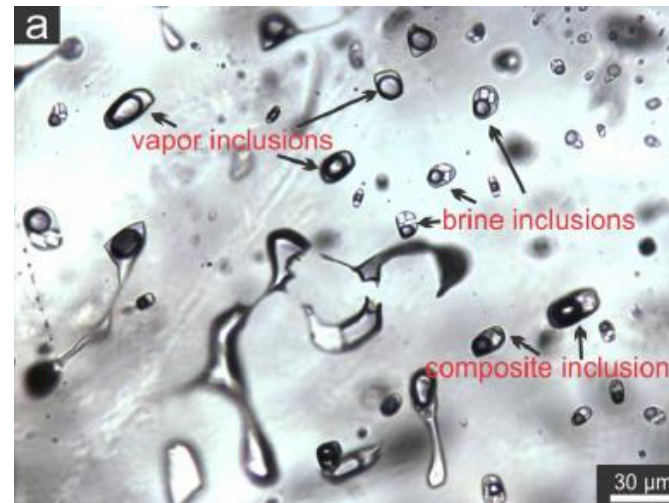
Batch-reactor / Autoclave

$T < 600\text{ }^{\circ}\text{C} - P < 200\text{ MPa}$



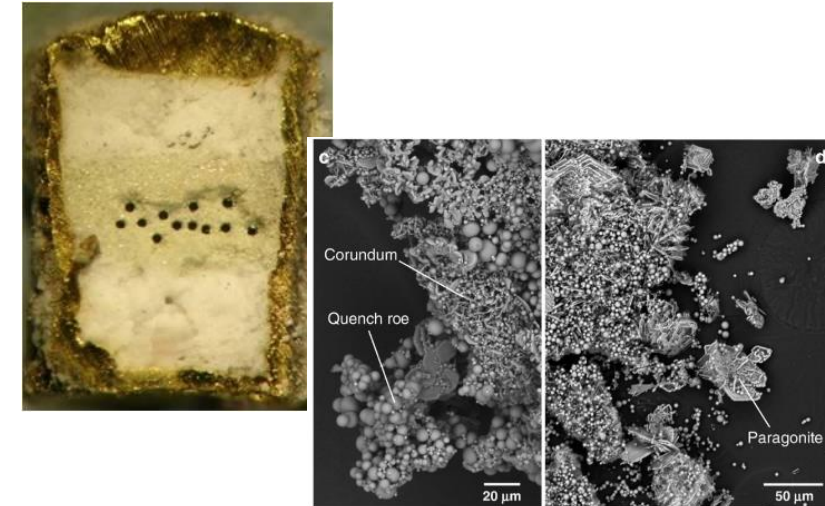
Synthetic fluid inclusions

$600 < T < 1000\text{ }^{\circ}\text{C} - P < 500\text{ MPa}$



Diamond-trap/weight loss exp.

$600 < T < 1400\text{ }^{\circ}\text{C} - 0.5 < P < 6\text{ GPa}$



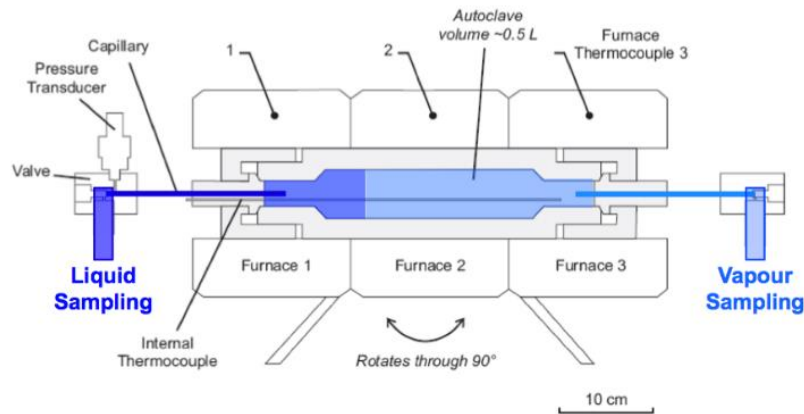
References: Antignano and Manning, 2008; Kessel et al., 2005; Dvir and Kessel, 2017 (*deep fluids equilibrated with eclogites*); Loges et al., 2013 (*YF3 solubility*); Pokrovski et al., 2005-2008 (*fluid-vapor partitioning of metals*); Zajacz et al., 2009-2017 (*Cu-Au-S solubility and fluid-melt-vapor partitioning*)

Challenging studies....

=> Experimental constraints are required !

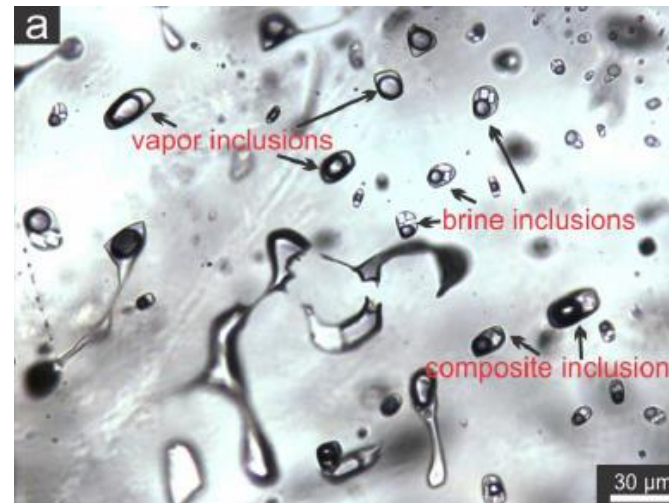
Batch-reactor / Autoclave

$T < 600\text{ }^{\circ}\text{C} - P < 200\text{ MPa}$



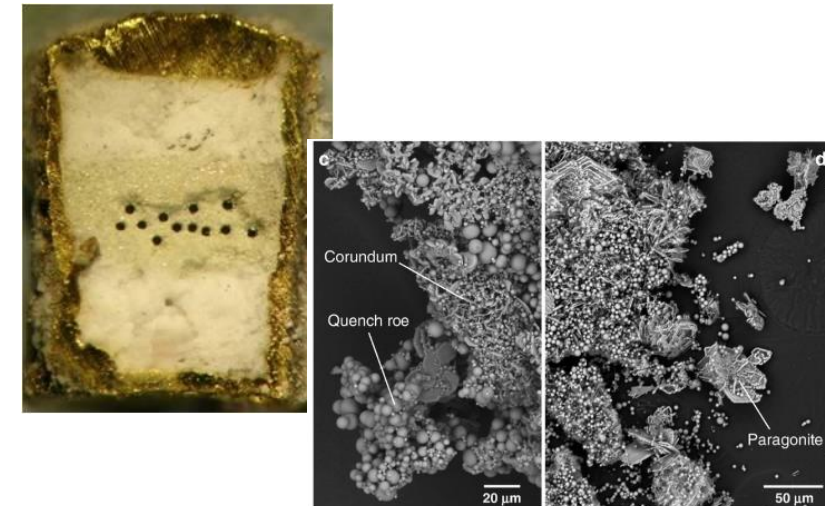
Synthetic fluid inclusions

$600 < T < 1000\text{ }^{\circ}\text{C} - P < 500\text{ MPa}$



Diamond-trap/weight loss exp.

$600 < T < 1400\text{ }^{\circ}\text{C} - 0.5 < P < 6\text{ GPa}$



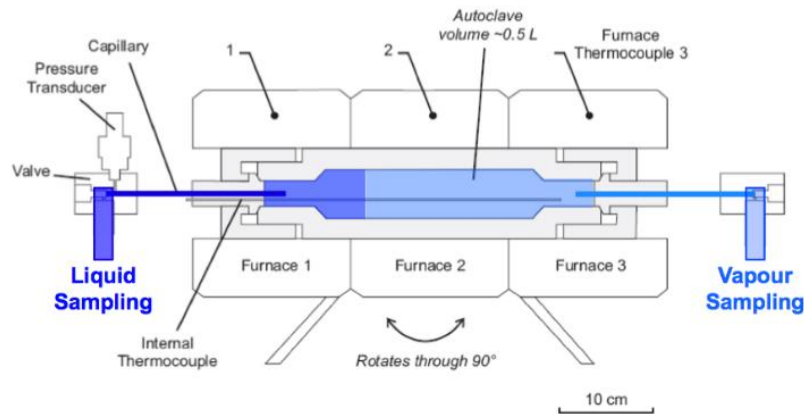
✓ Great for quantification

Challenging studies....

=> Experimental constraints are required !

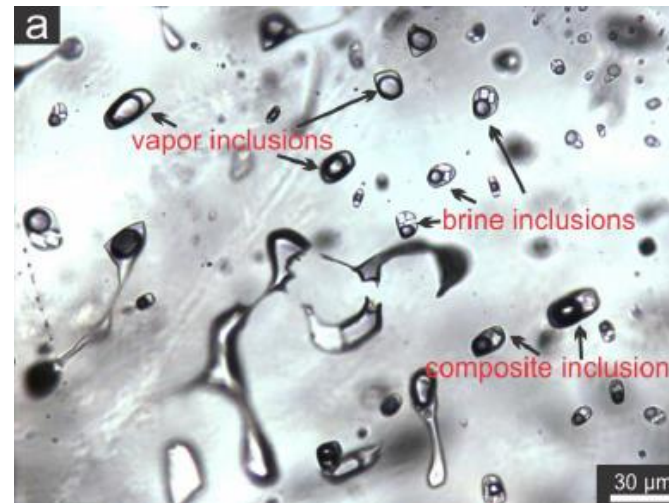
Batch-reactor / Autoclave

$T < 600\text{ }^{\circ}\text{C} - P < 200\text{ MPa}$



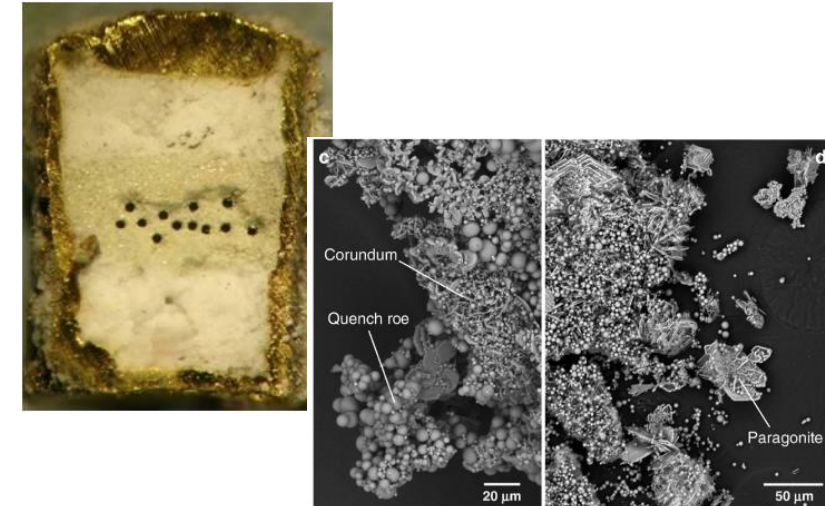
Synthetic fluid inclusions

$600 < T < 1000\text{ }^{\circ}\text{C} - P < 500\text{ MPa}$



Diamond-trap/weight loss exp.

$600 < T < 1400\text{ }^{\circ}\text{C} - 0.5 < P < 6\text{ GPa}$



- ✓ Great for quantification
- ✗ High P-T species unknown
- ✗ Nature of high T fluids (brine, gas)?

In-situ spectroscopy on high T fluids

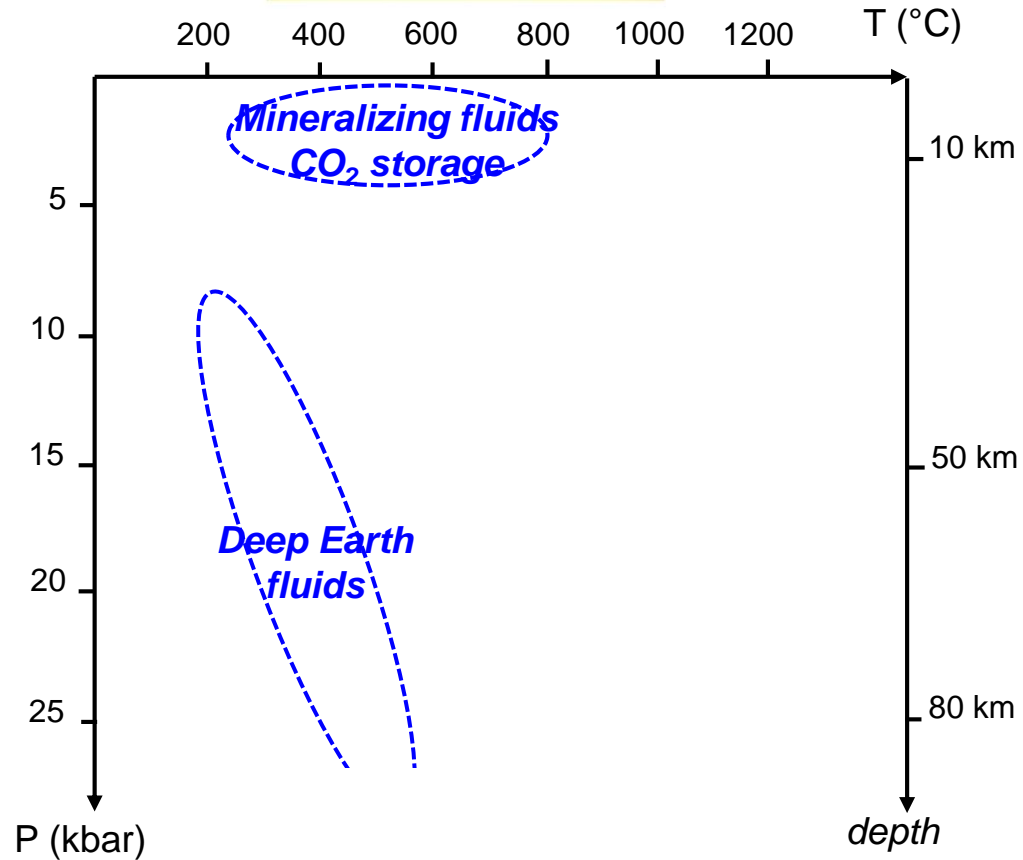
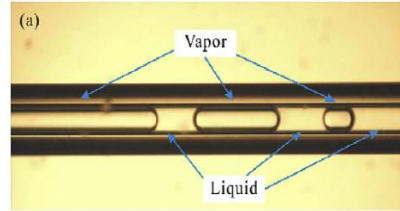
Autoclave HT

600 °C – 0.1-2kbar



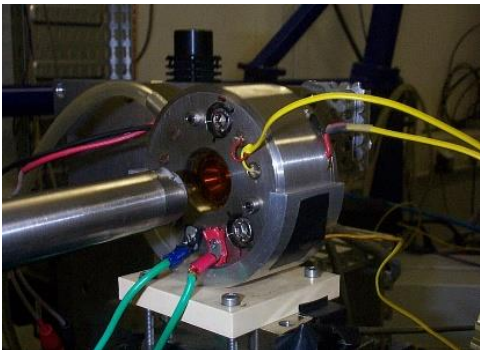
Si fused capillaries

~ 500 °C - P_{sat}



HDAC

900 °C – 5-30 kbar



In-situ spectroscopy on high T fluids

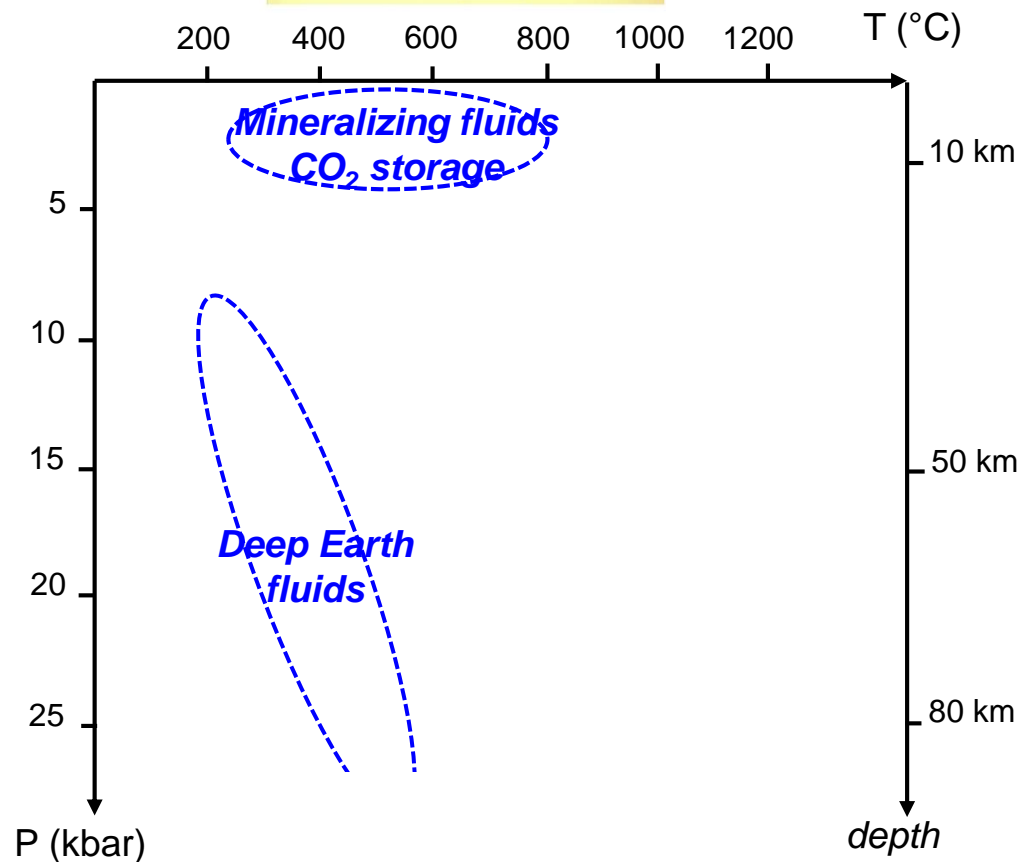
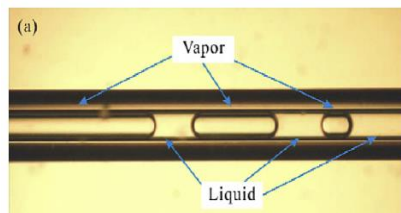
Autoclave HT

600 °C – 0.1-2kbar



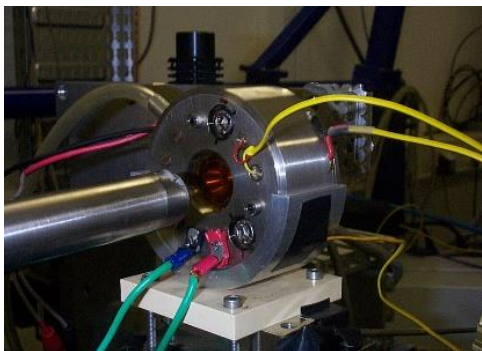
Si fused capillaries

~ 500 °C - P_{sat}



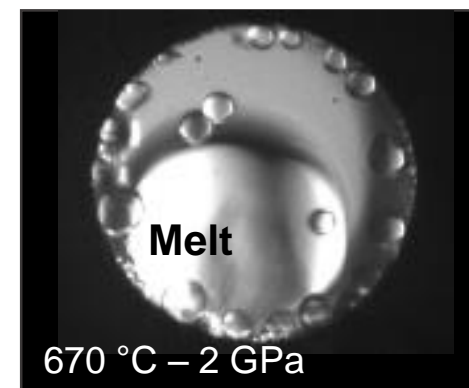
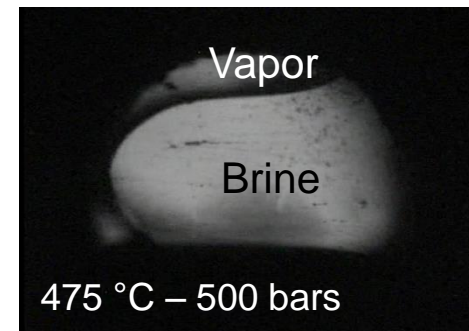
HDAC

900 °C – 5-30 kbar



1. Visualization

phase separation, homogeneization, melting



In-situ spectroscopy on high T fluids

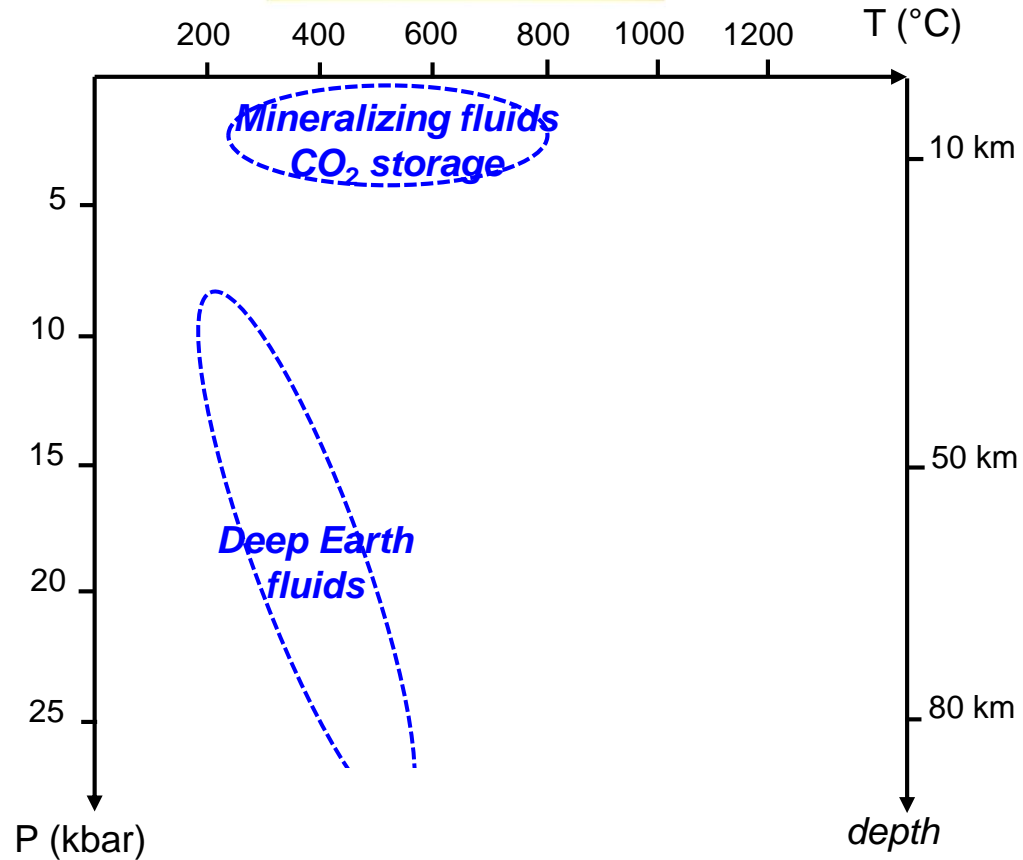
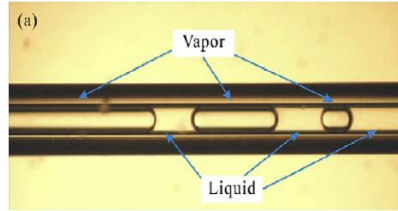
Autoclave HT

600 °C – 0.1-2kbar



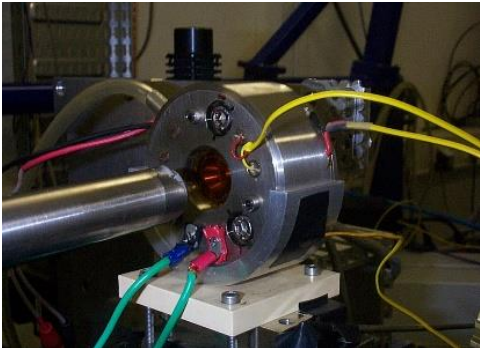
Si fused capillaries

~ 500 °C - P_{sat}



HDAC

900 °C – 5-30 kbar

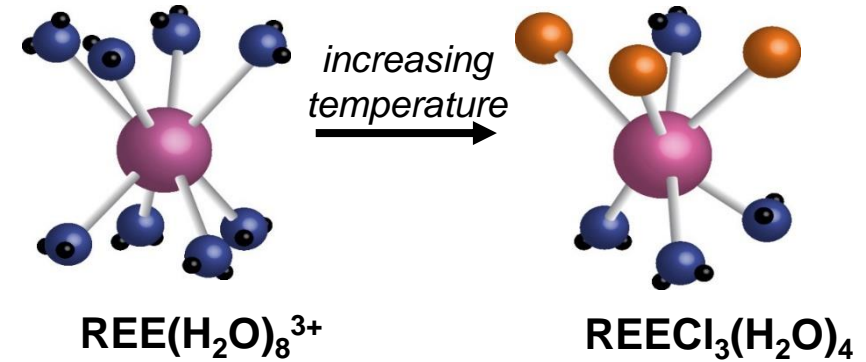


1. Visualization

phase separation, homogenization, melting

2. X-ray absorption/fluorescence

solubility, speciation, density



In-situ spectroscopy on high T fluids

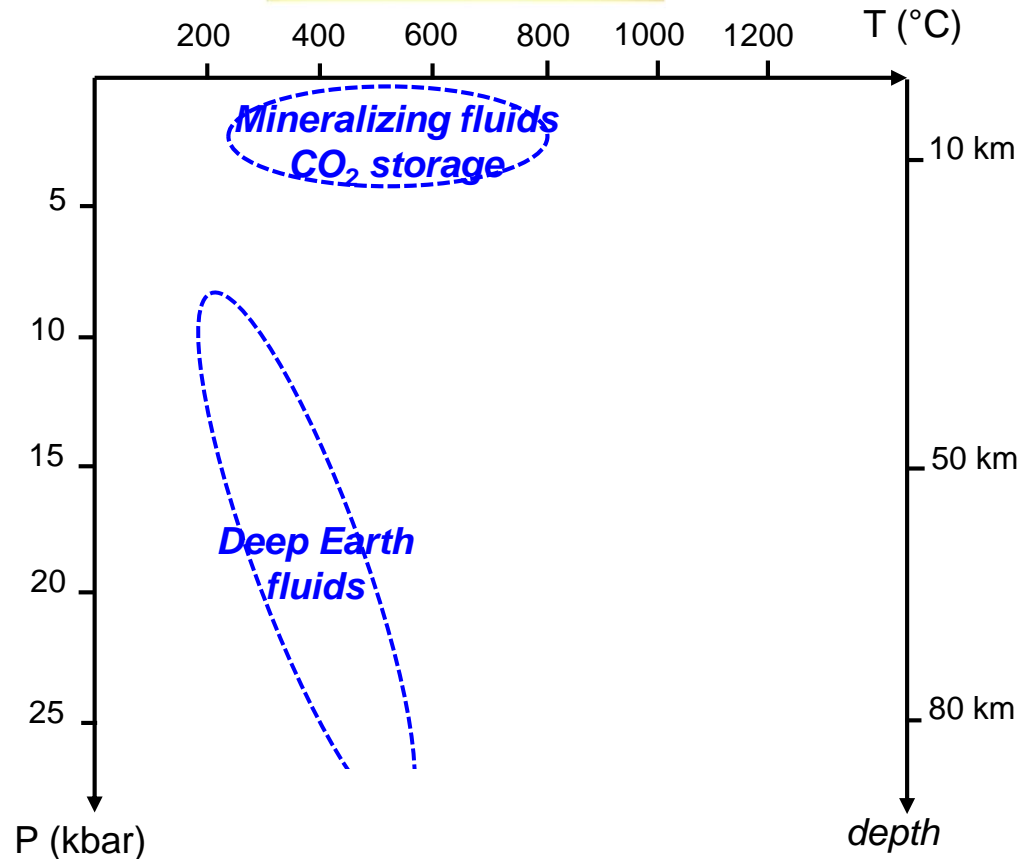
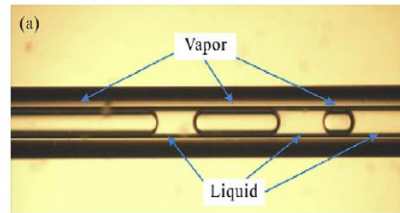
Autoclave HT

600 °C – 0.1-2kbar



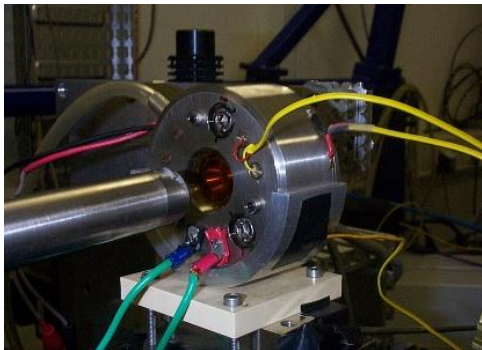
Si fused capillaries

~ 500 °C - P_{sat}



HDAC

900 °C – 5-30 kbar



1. Visualization

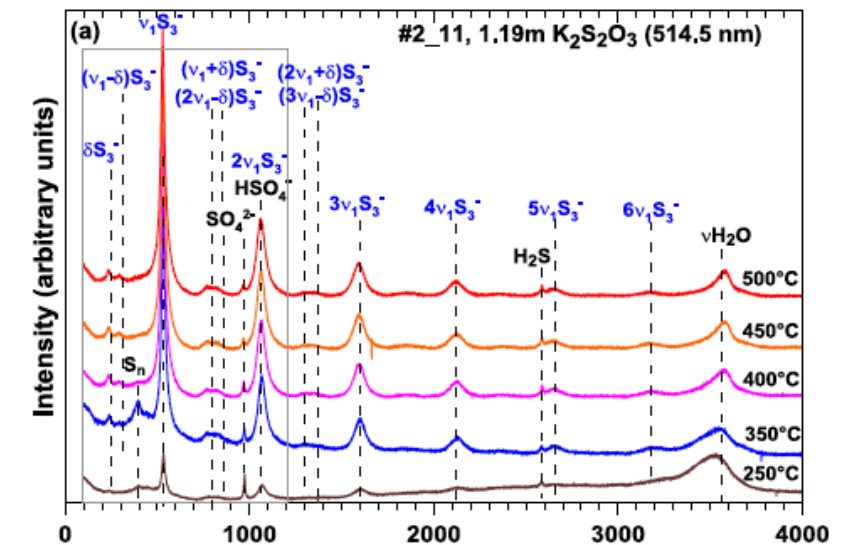
phase separation, homogenization, melting

2. X-ray absorption/fluorescence

solubility, speciation, density

3. Raman

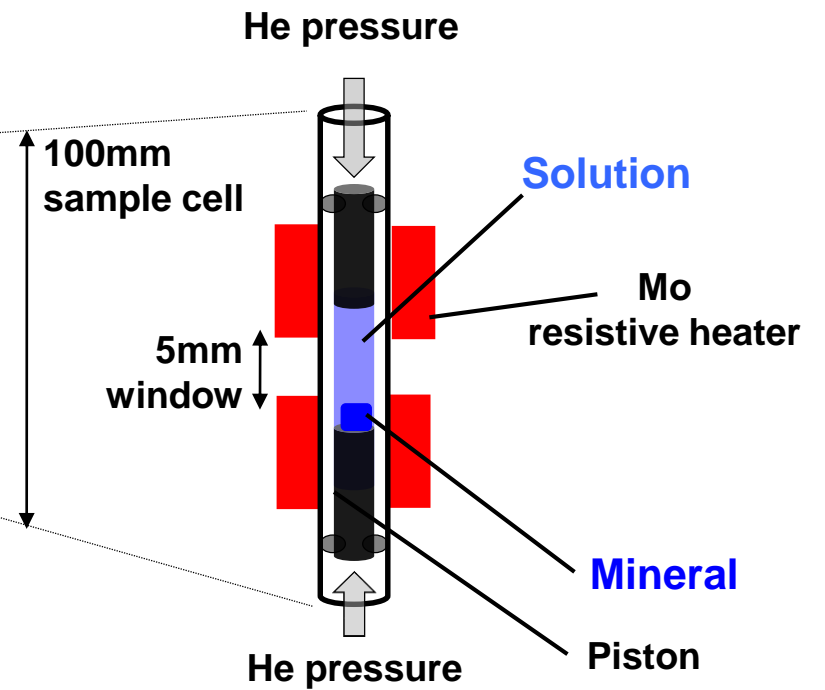
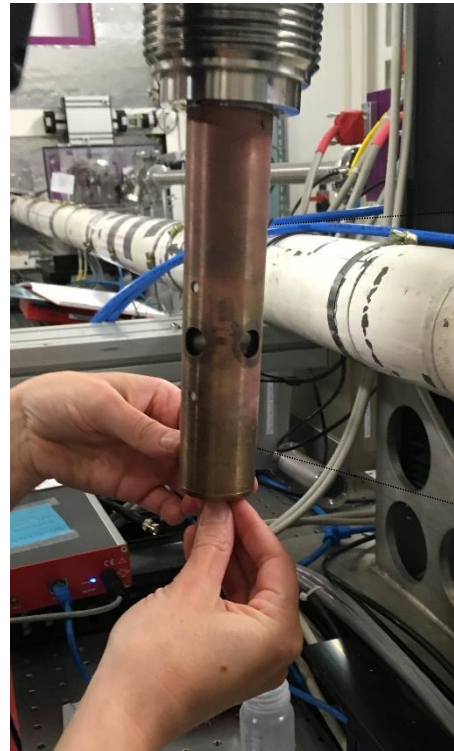
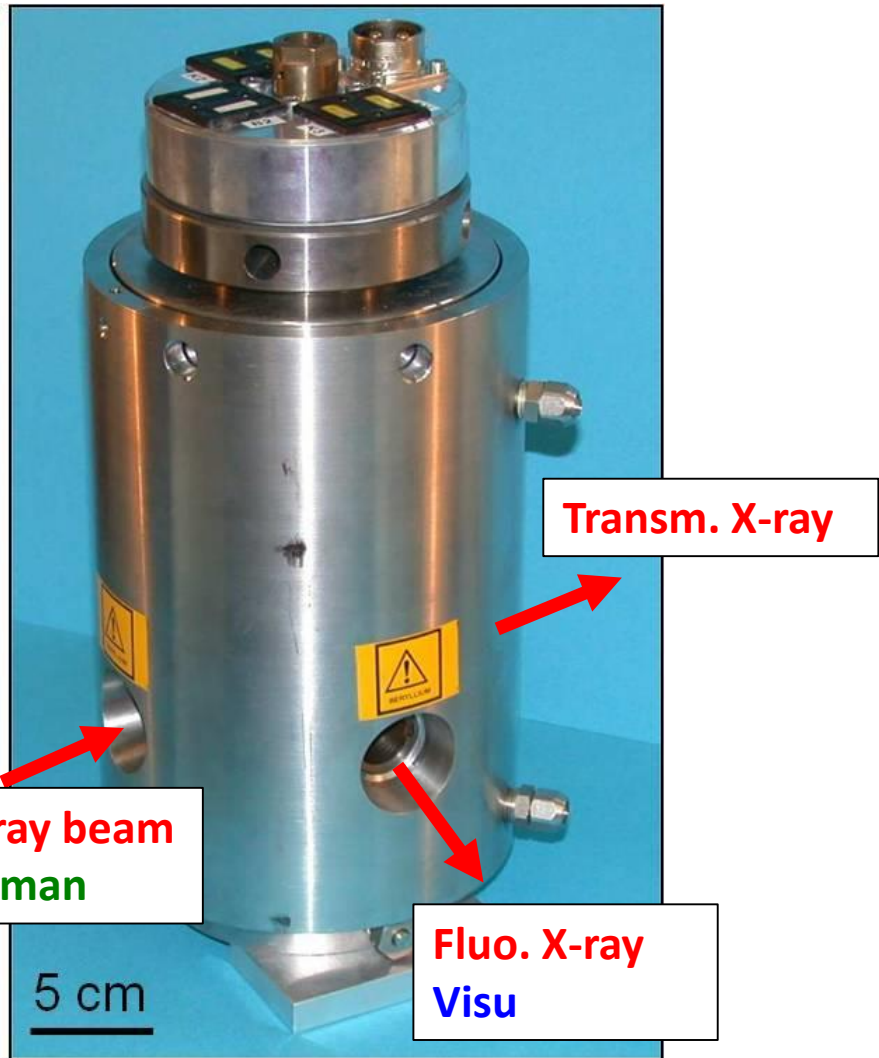
volatile species, structure of solvent



Pokrovski et al., 2011-2015

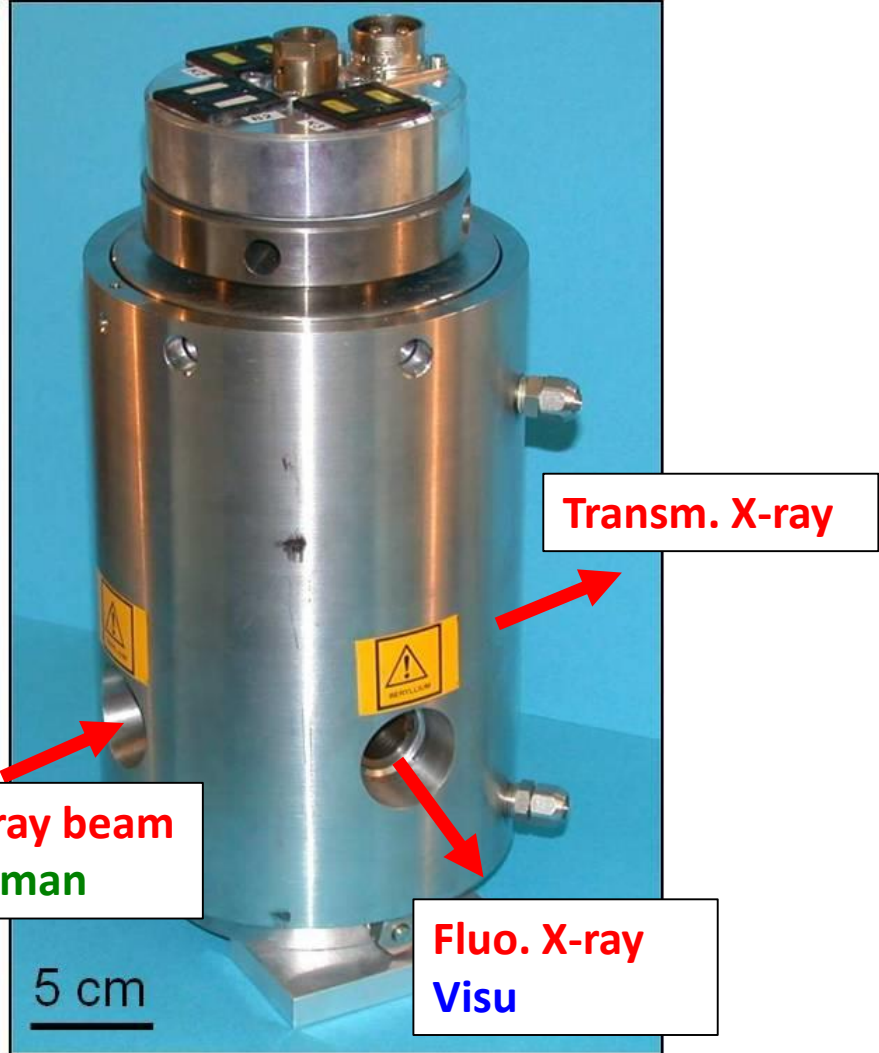
The autoclave

The autoclave : Concept and set-up for *in-situ* spectroscopy



Pressure ± 0.4 bars
Temperature ± 0.1 °C

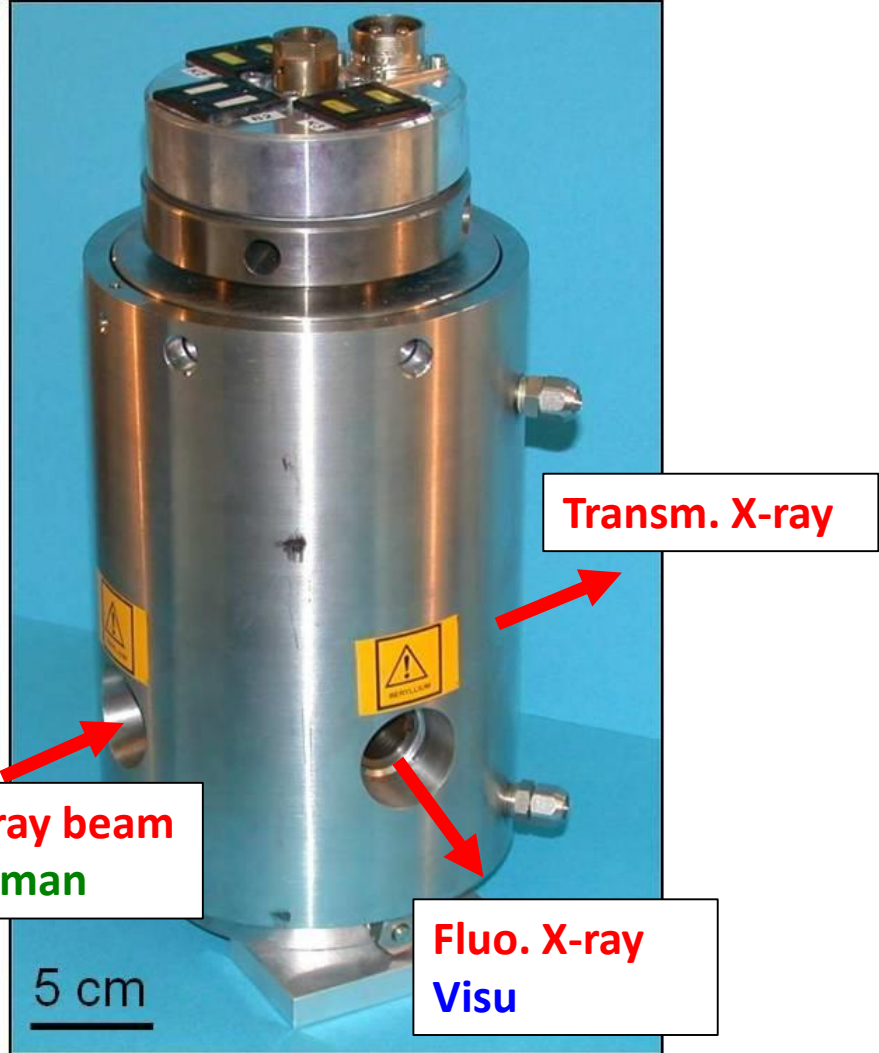
The autoclave : Concept and set-up for *in-situ* spectroscopy



Heater : Cu or Mo



The autoclave : Concept and set-up for *in-situ* spectroscopy



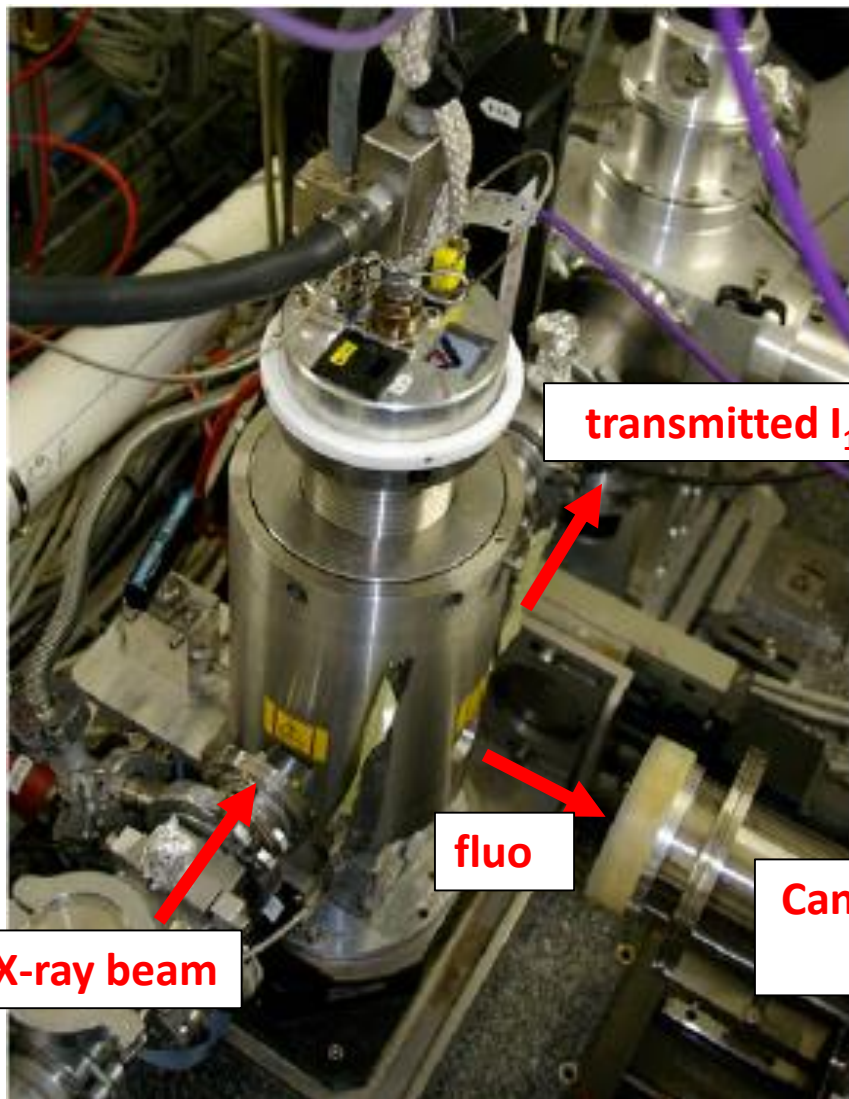
Heater : Cu or Mo



Autoclave I : in-situ X-ray absorption

The autoclave : Concept and set-up for *in-situ* spectroscopy

➤ *in-situ* XAS

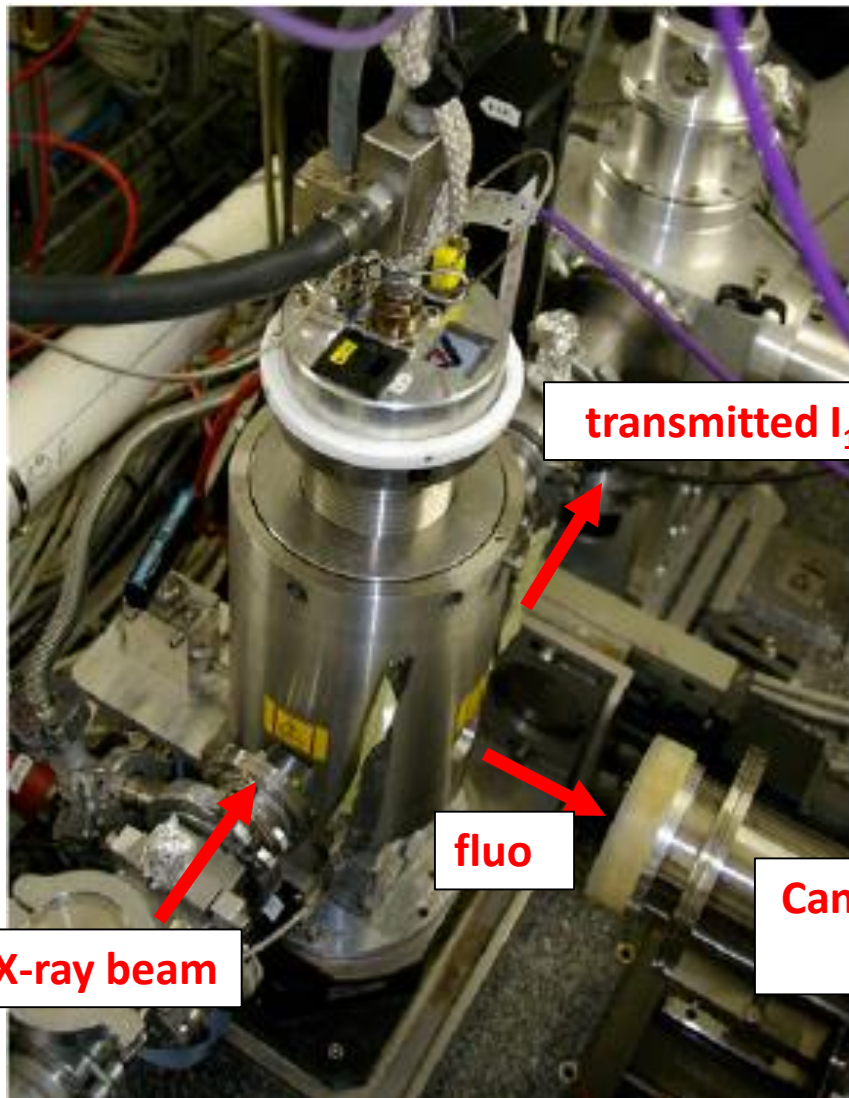


Canberra 30 elmt
detector



The autoclave : Concept and set-up for *in-situ* spectroscopy

➤ *in-situ* XAS



transmitted I_1

fluo

Canberra 30 elmt
detector

X-ray beam

Be or C windows

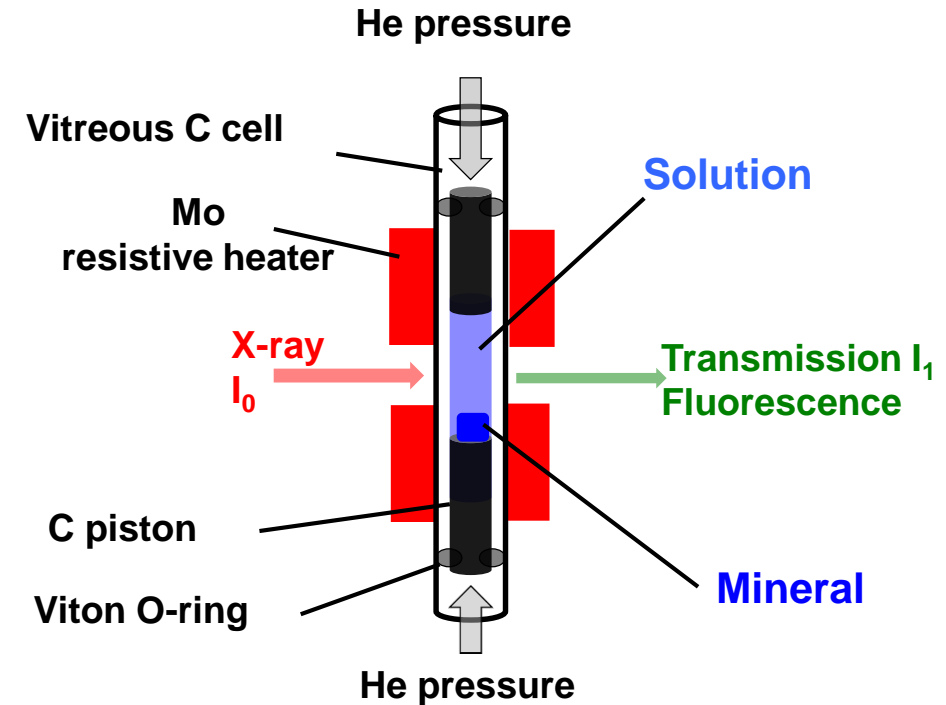
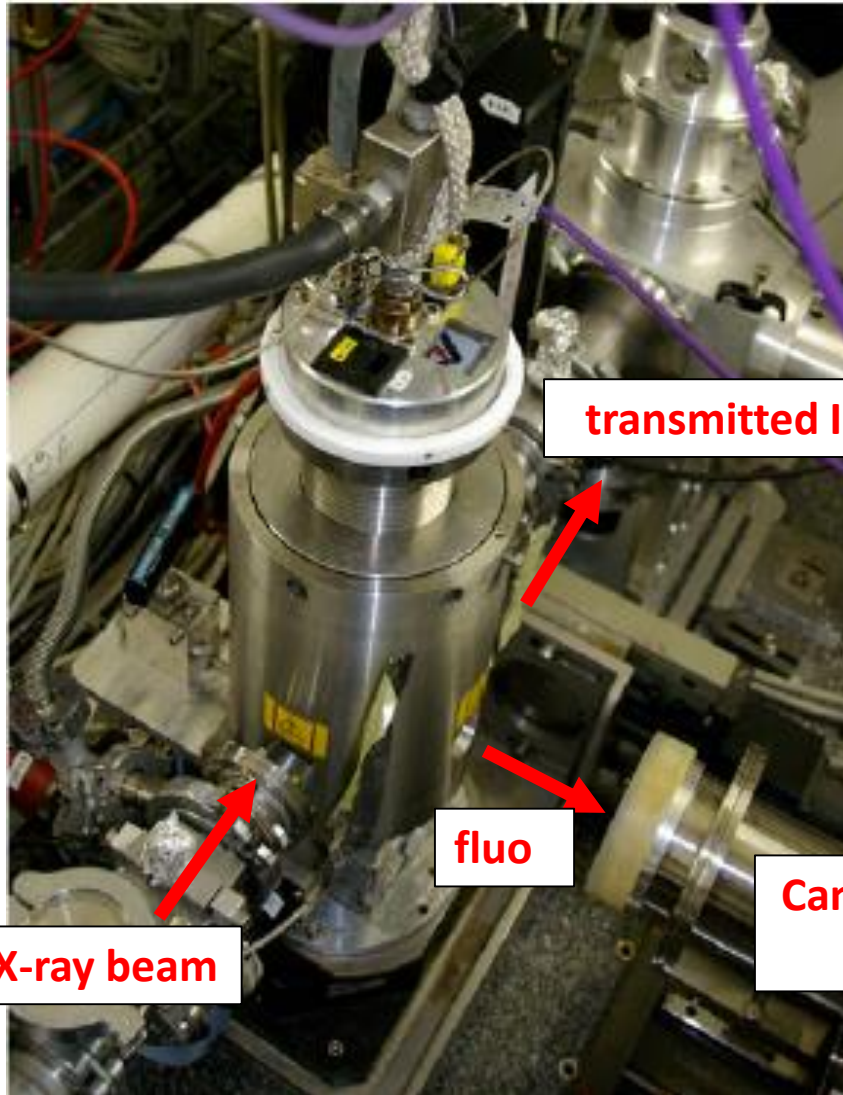


Vitreous C sample container



The autoclave : Concept and set-up for *in-situ* spectroscopy

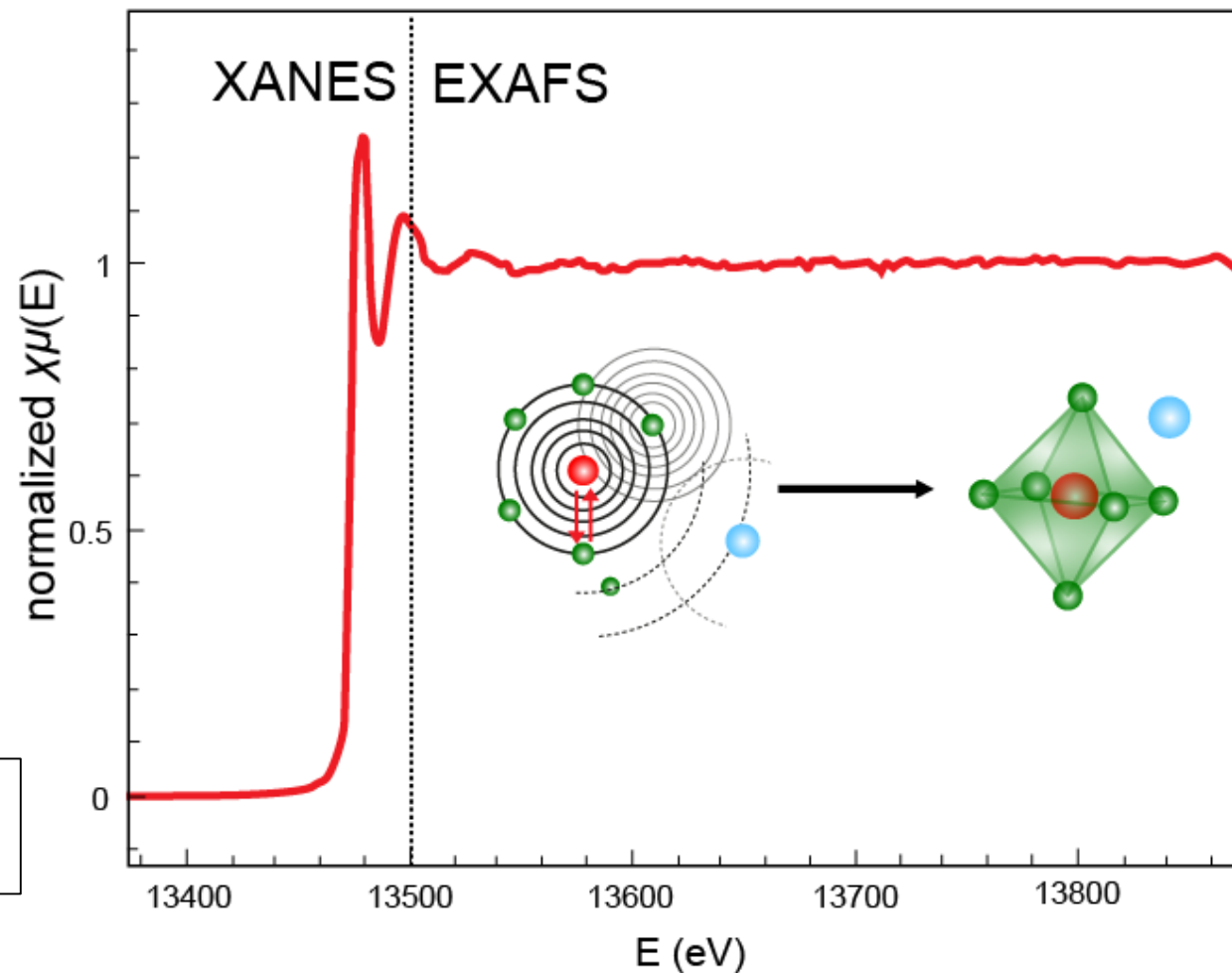
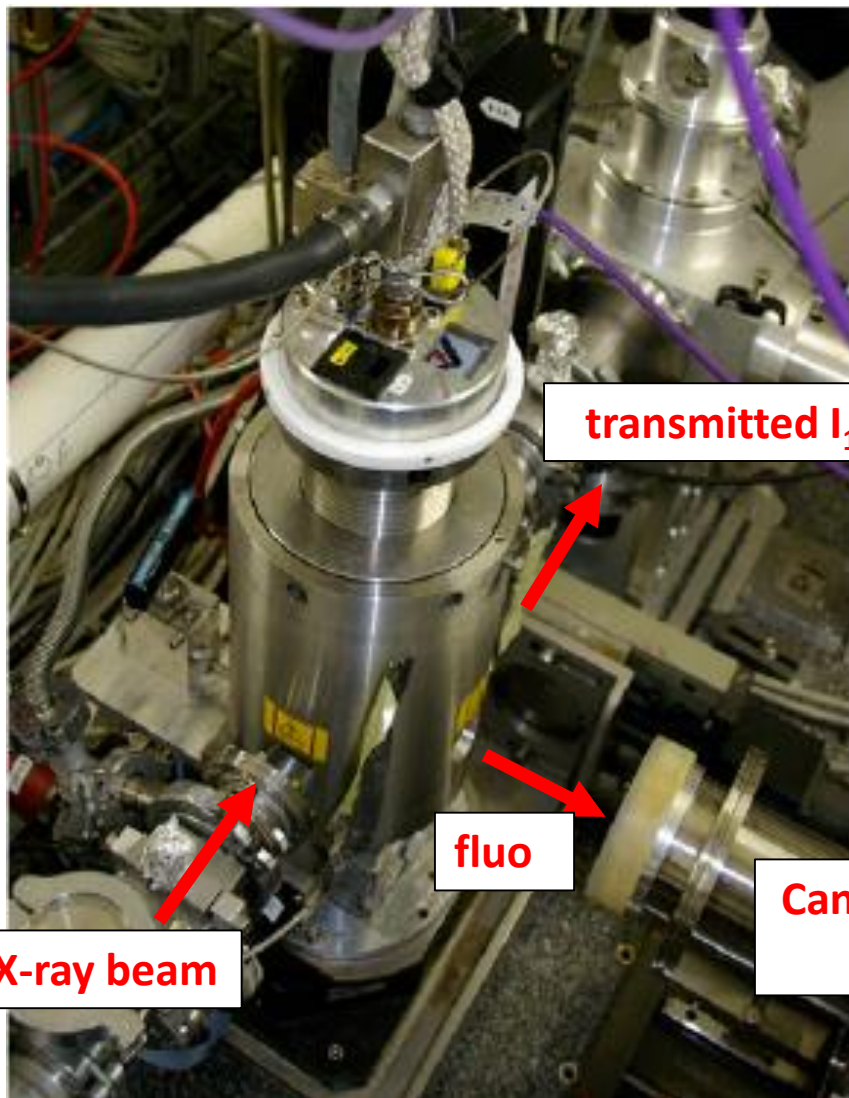
➤ *in-situ* XAS



- **Experimental Conditions:**
Beam ~ 200x100 μm HxV (FWHM)
P is known and T calibrated to water EOS
6 < E < 25 keV (Mn K-edge/La L3-edge ; Sb K-edge)

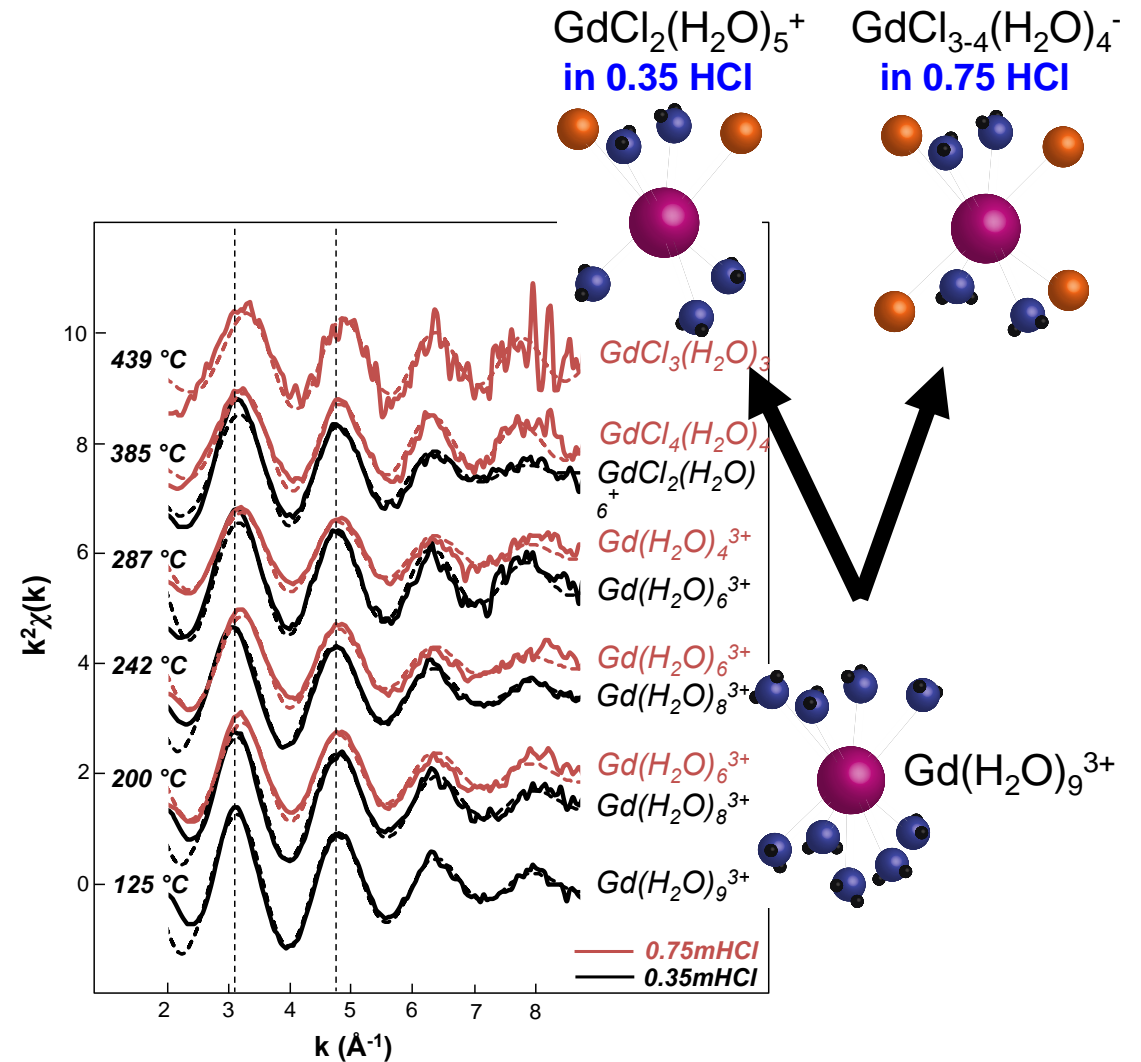
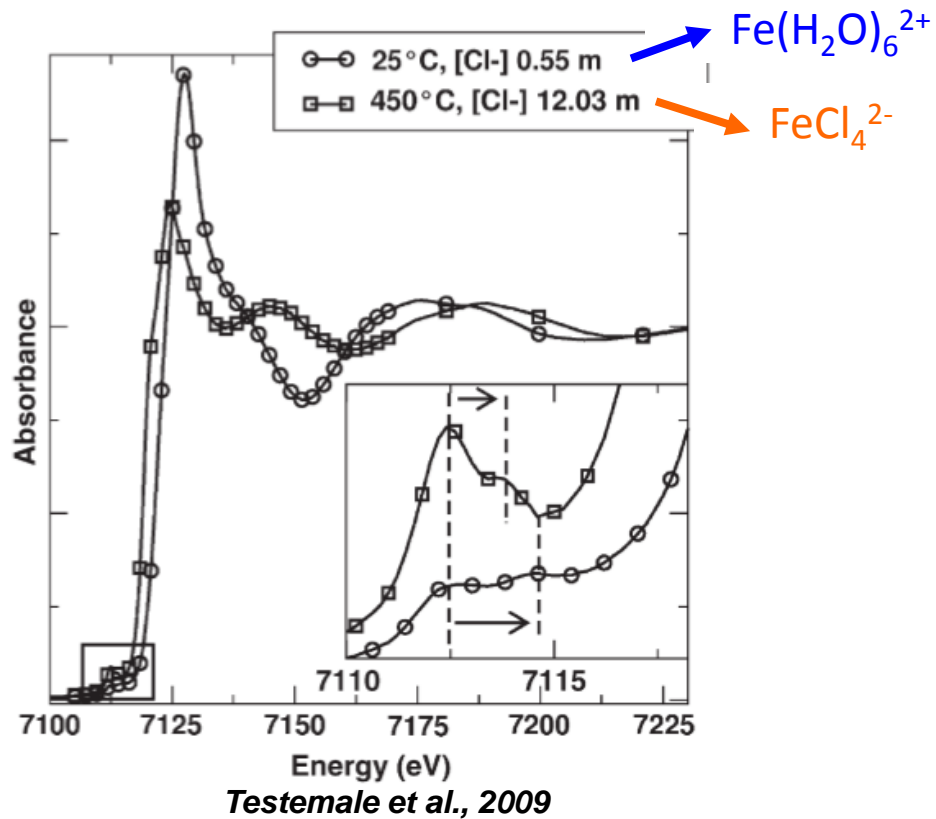
The autoclave : Concept and set-up for *in-situ* spectroscopy

➤ *in-situ* XAS : Speciation (XANES + EXAFS)



The autoclave : Concept and set-up for *in-situ* spectroscopy

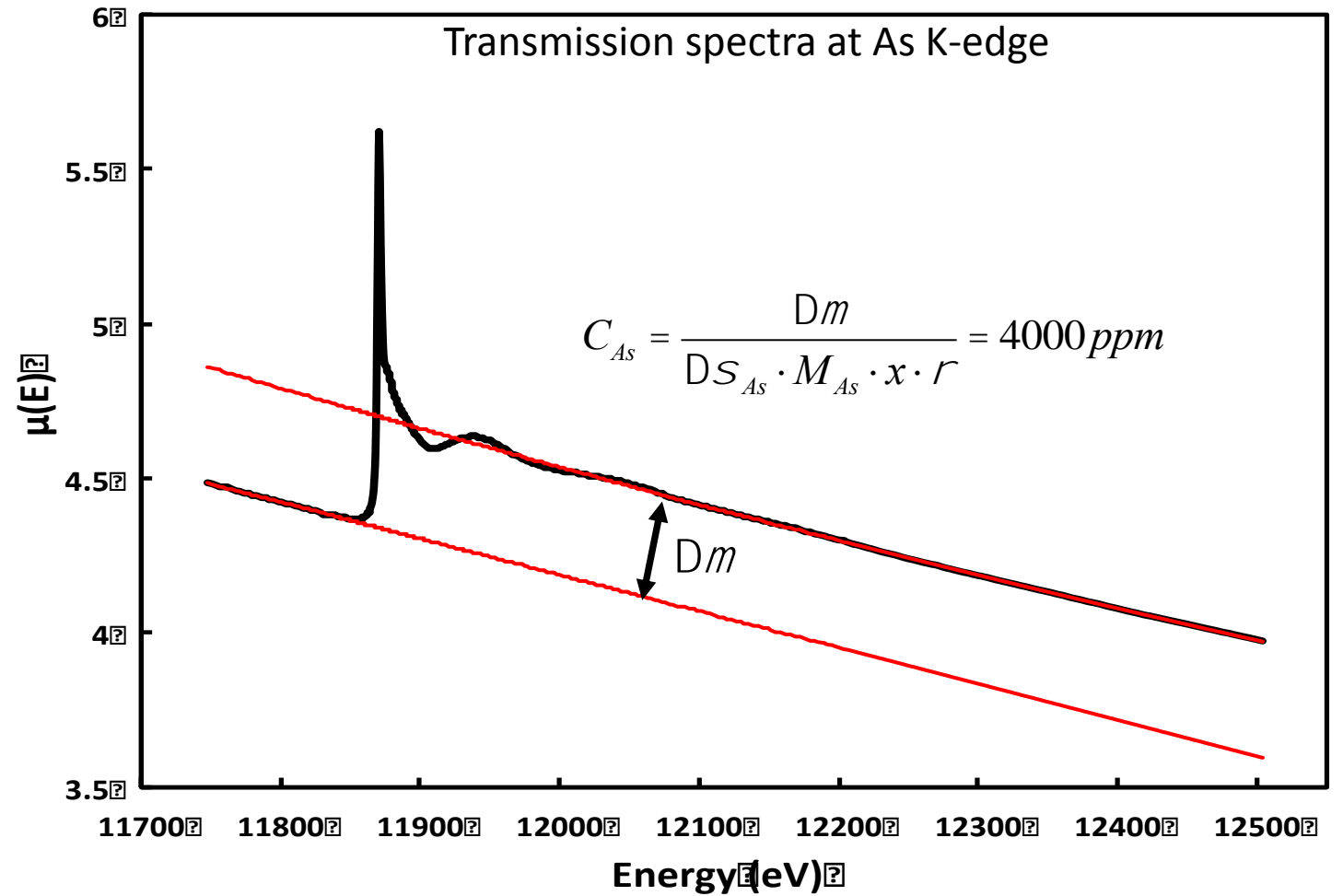
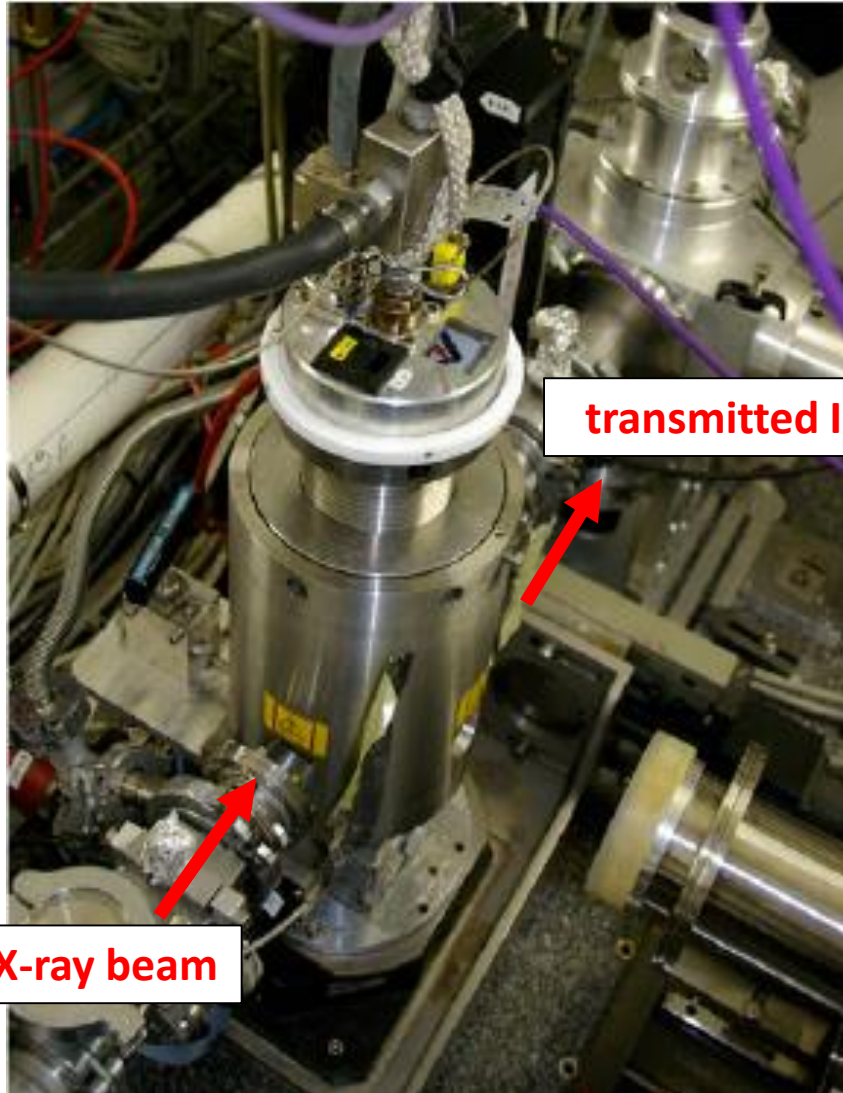
➤ *in-situ* XAS : Speciation (XANES + EXAFS)



References: Pokrovski et al, 2005, 2009, 2013 (Ag, Au, As, Ge) Brugger, Etschmann, Liu et al., 2008, 2011, 2013, 2016, 2018 (Cu, Au, Zn, Co, Pb, Bi, Eu); Bazarkina et al., 2010, 2014 (Cd, Pd); Dargent et al, 2013 (U); Louvel et al., 2015, 2017 (Cu, Yb); Testemale et al., 2009 (Fe)

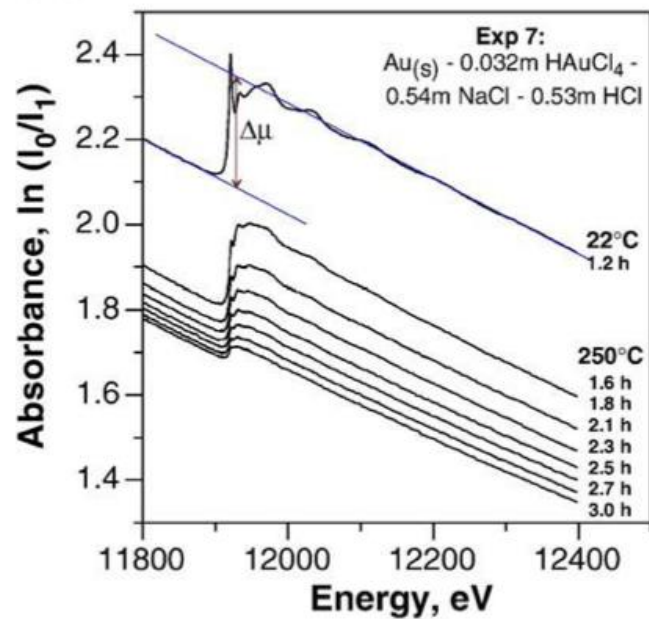
The autoclave : Concept and set-up for *in-situ* spectroscopy

➤ *in-situ* XAS : Solubility

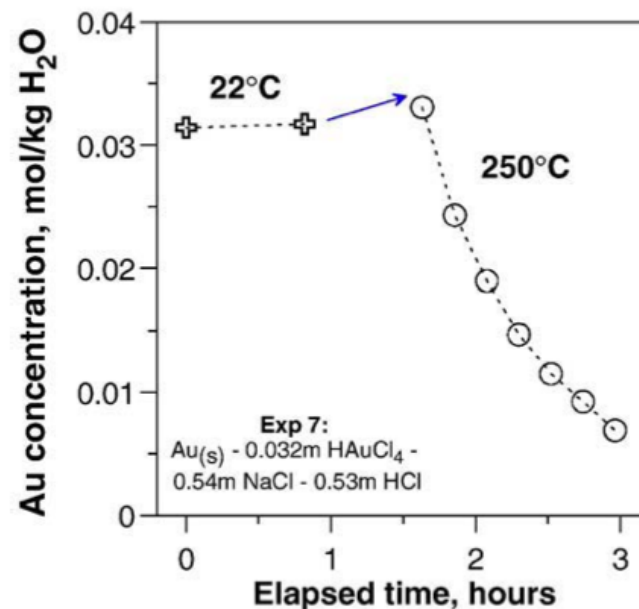


The autoclave : Concept and set-up for *in-situ* spectroscopy

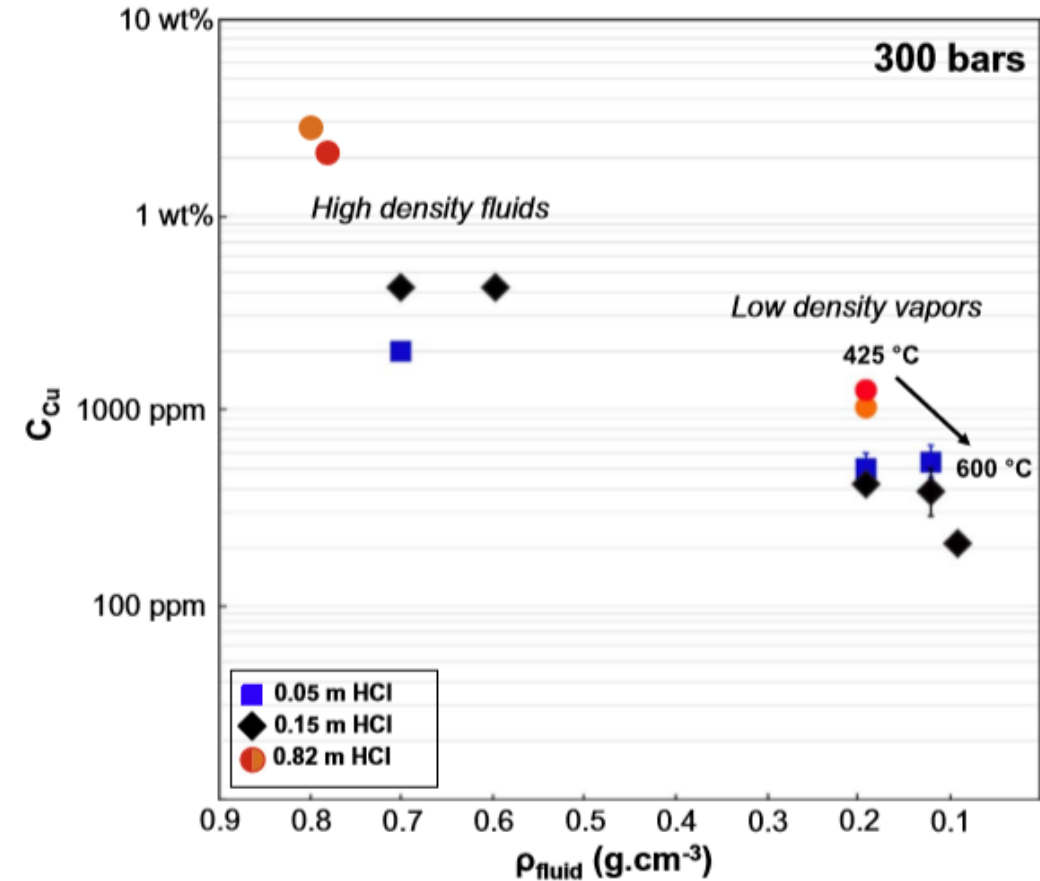
➤ *in-situ* XAS : Solubility



Au precipitation
in Au-NaCl-HCl fluids
Pokrovski et al., 2009

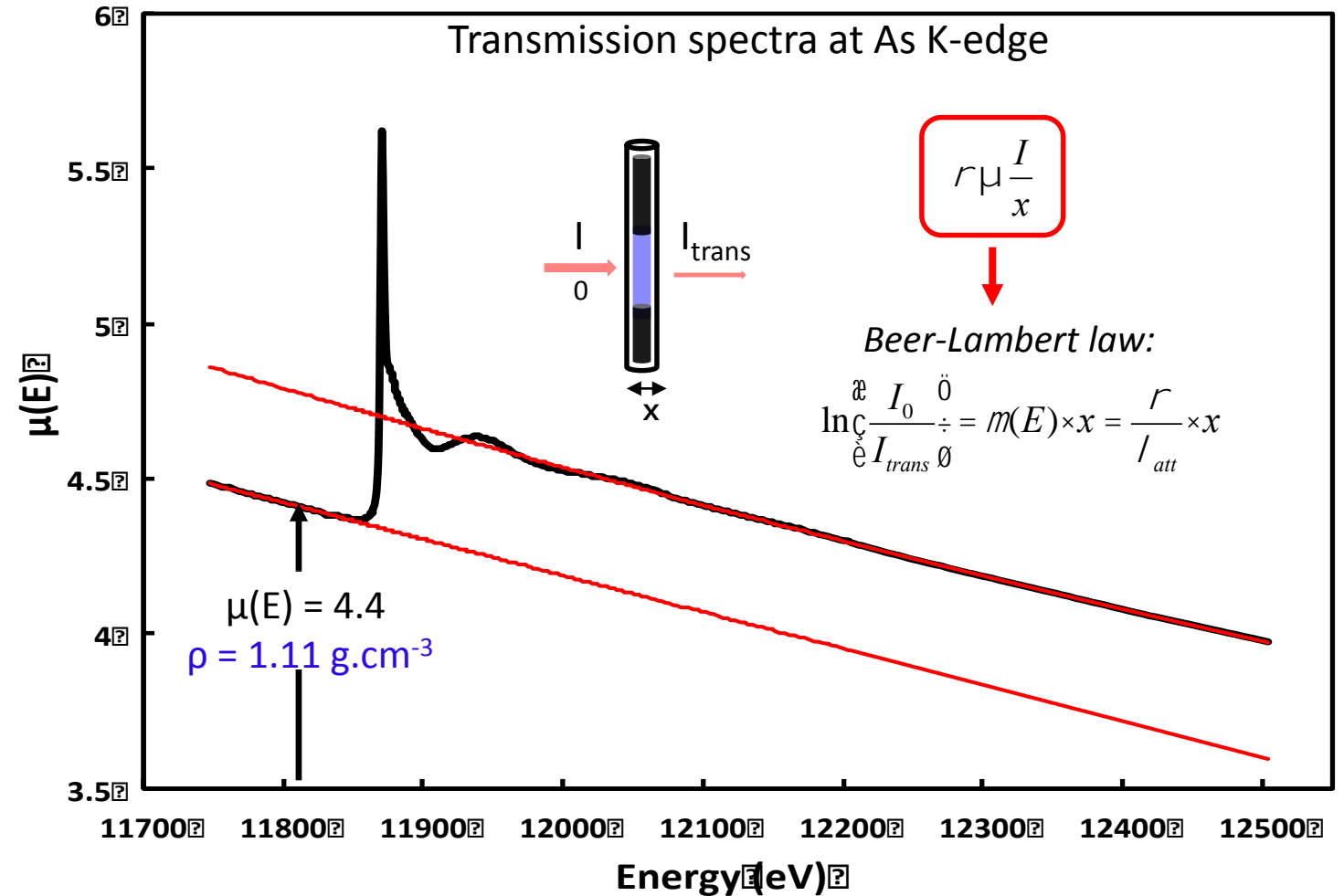
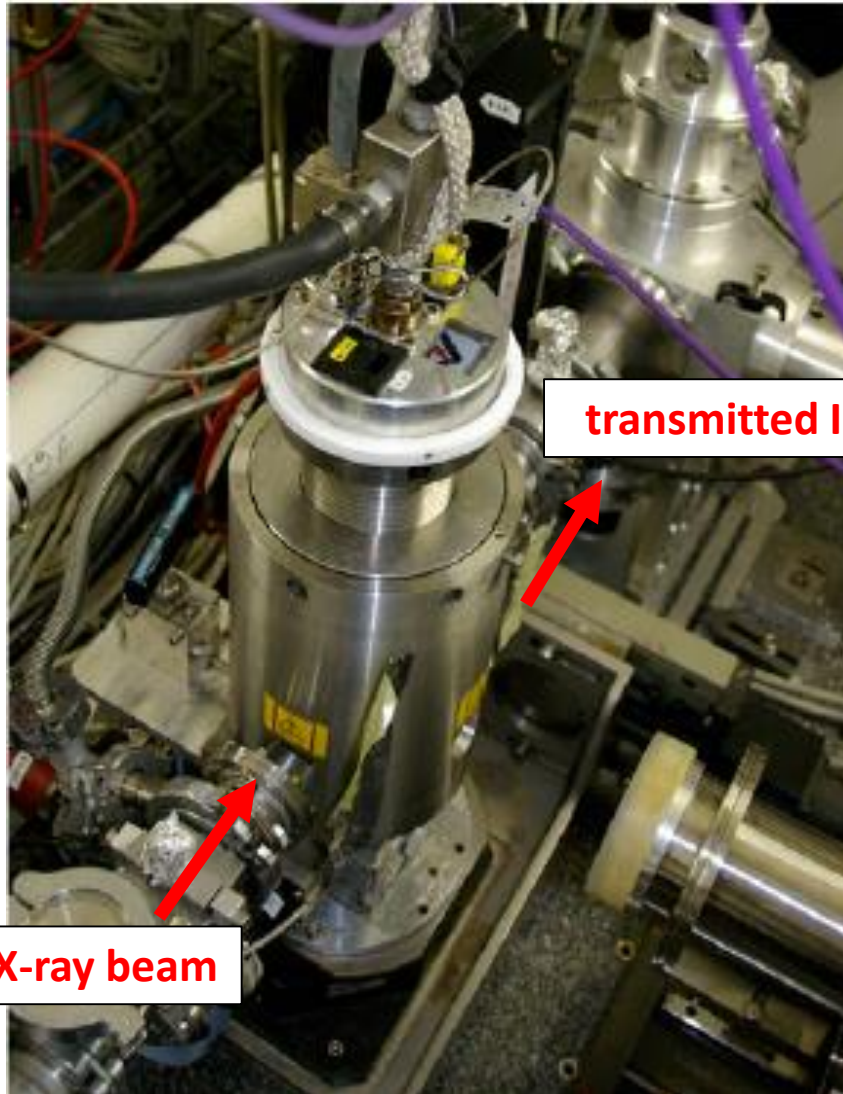


Cu solubility in HCl fluids and 'gas'
Louvel et al., 2017



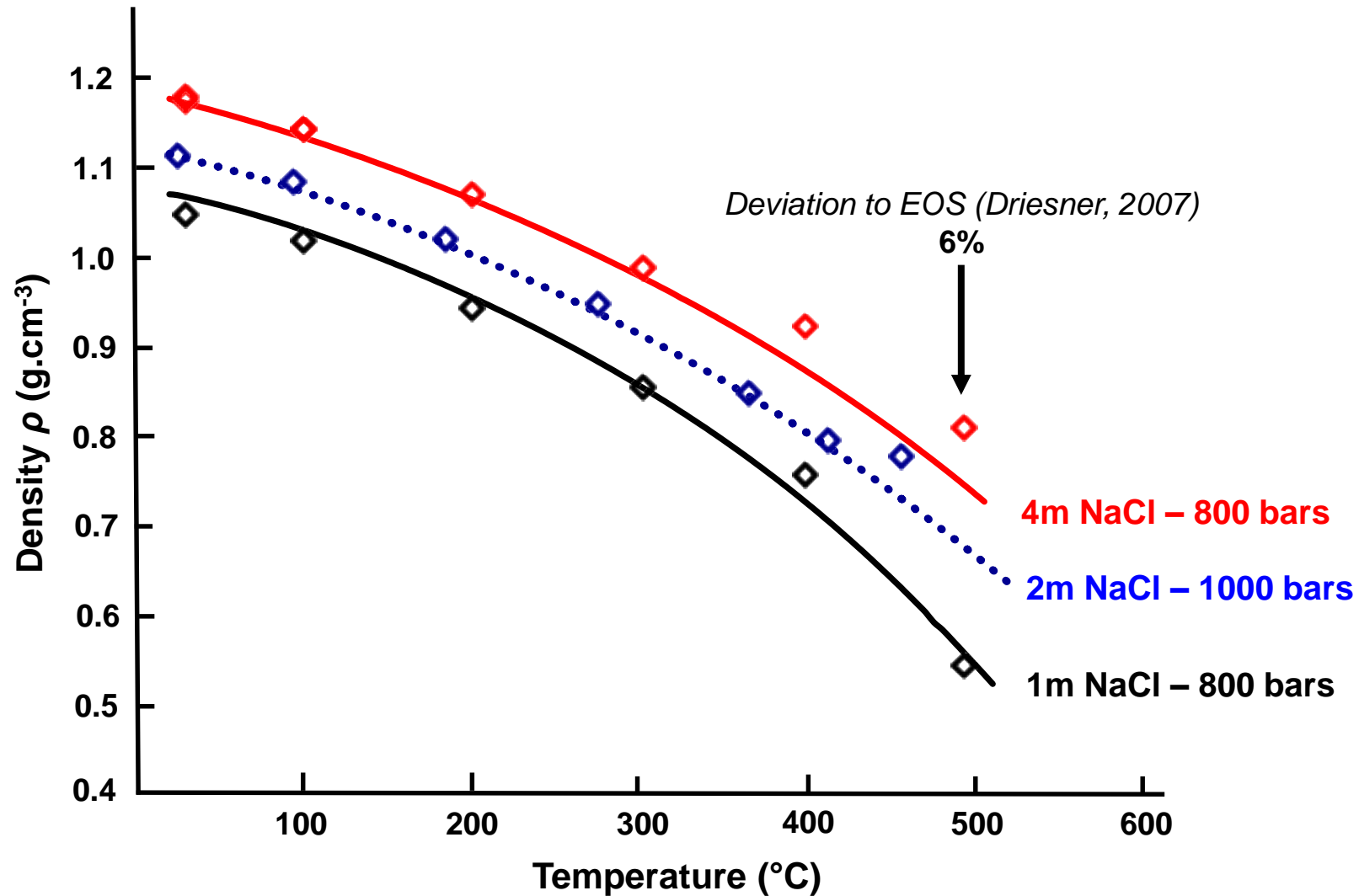
The autoclave : Concept and set-up for *in-situ* spectroscopy

➤ *in-situ* XAS : Density measurements



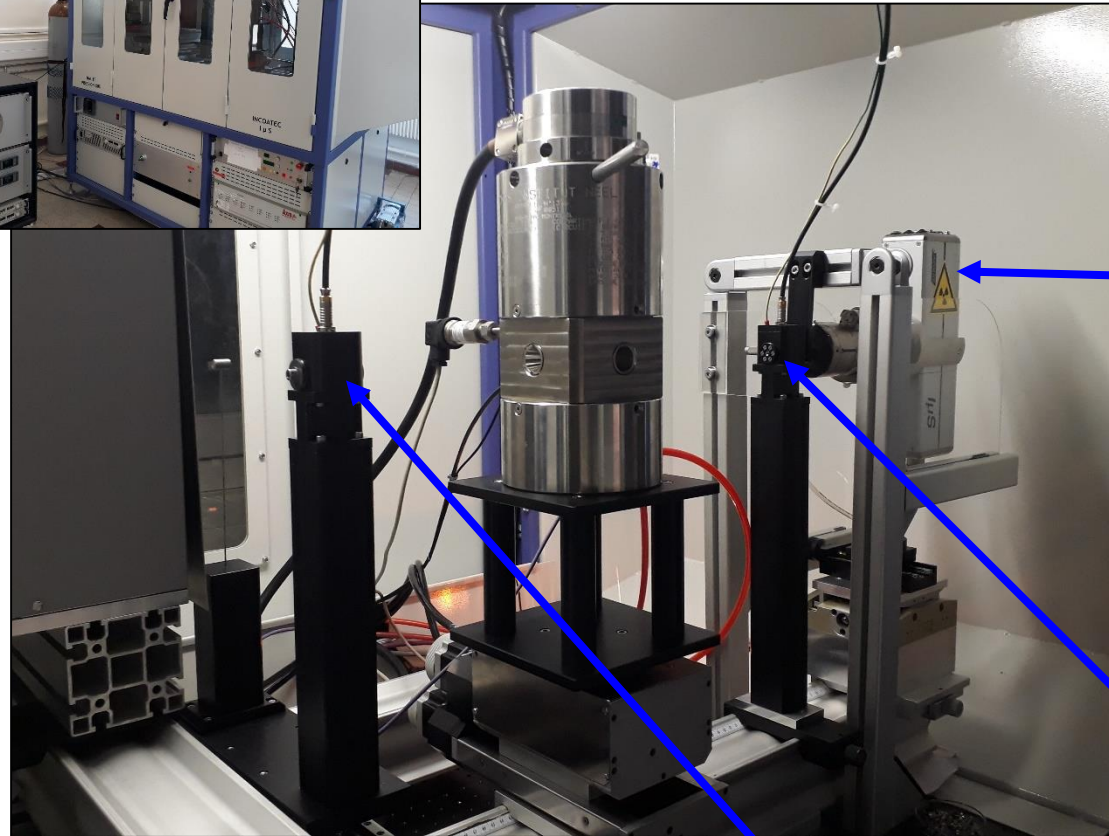
The autoclave : Concept and set-up for *in-situ* spectroscopy

➤ *in-situ* XAS : Density measurements



The autoclave : Concept and set-up for *in-situ* spectroscopy

- *in-situ* XAS : Density measurements **without synchrotron at Institut Néel**



RX Incoatec Ag Source
22 keV
 $\approx 160 \mu\text{m}^2$

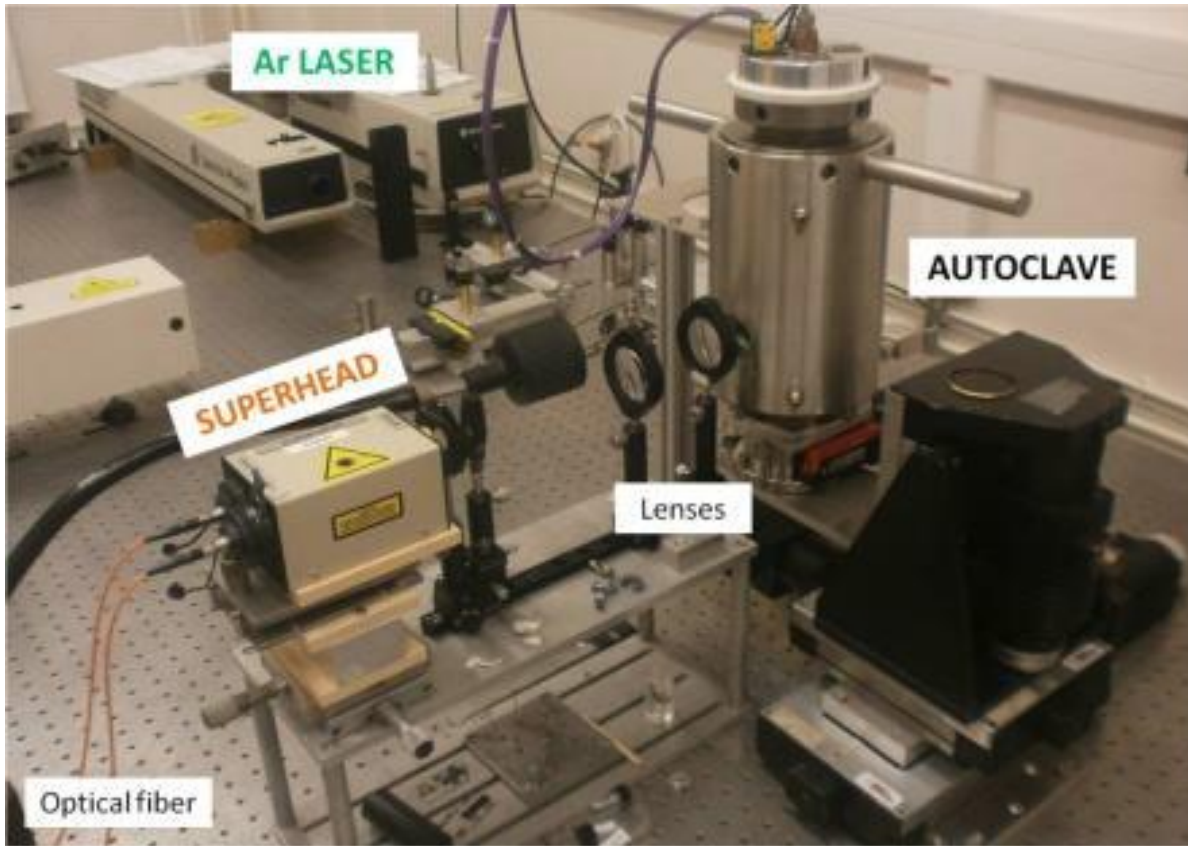
Beam I_0

Transmission I_1

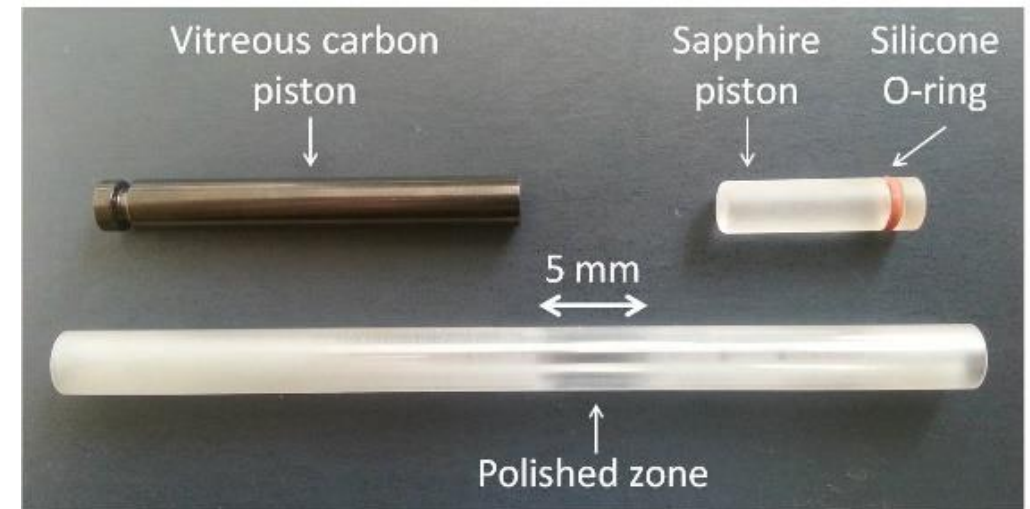
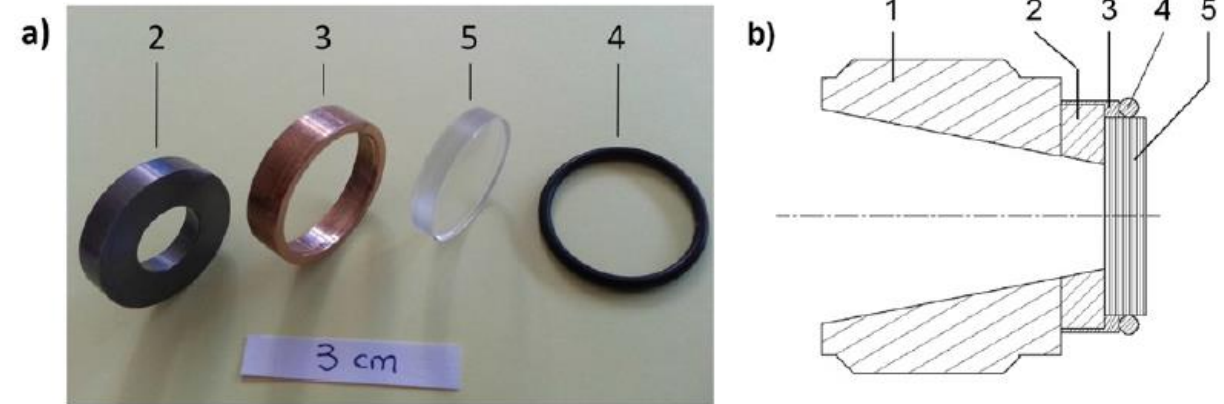
Autoclave II : in-situ Raman

The autoclave : Concept and set-up for *in-situ* spectroscopy

➤ *in-situ* Raman



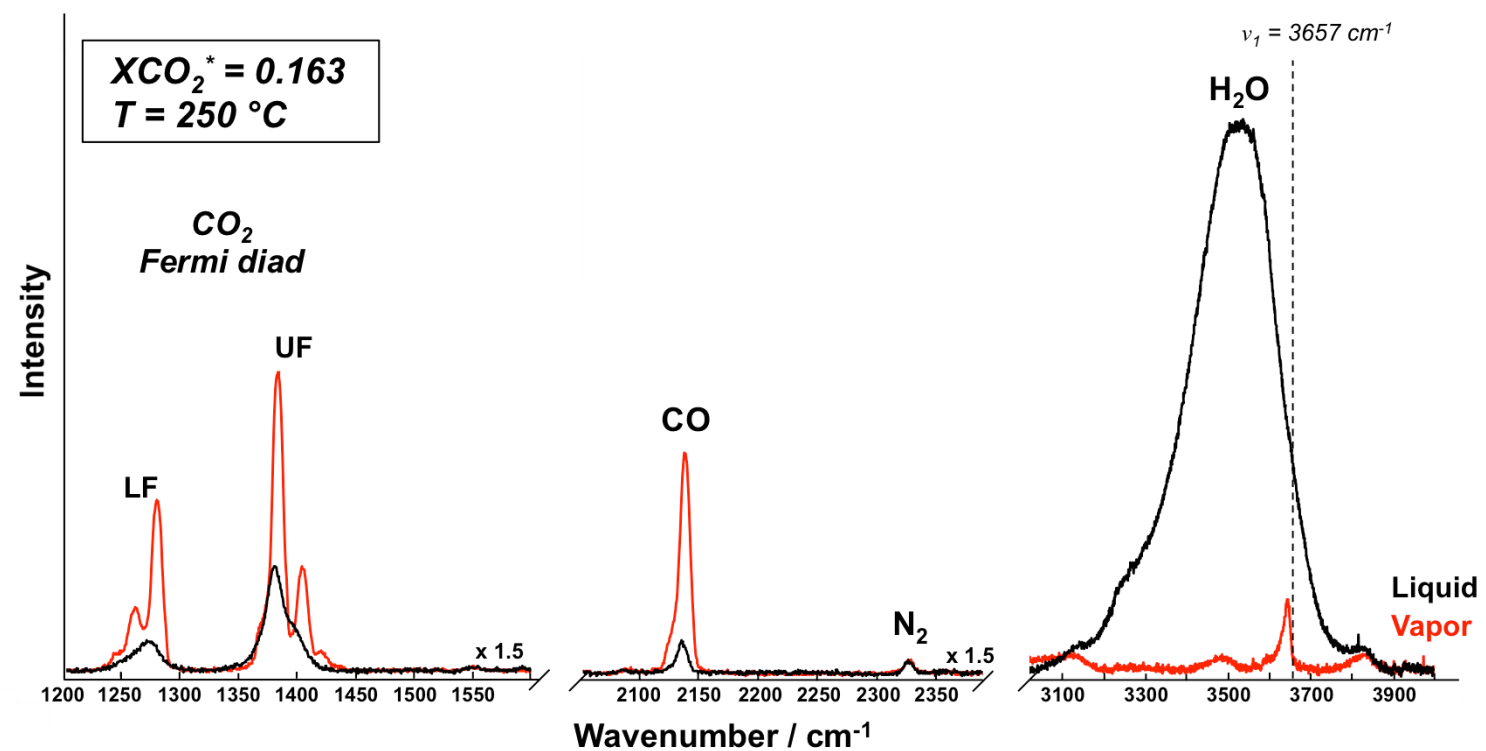
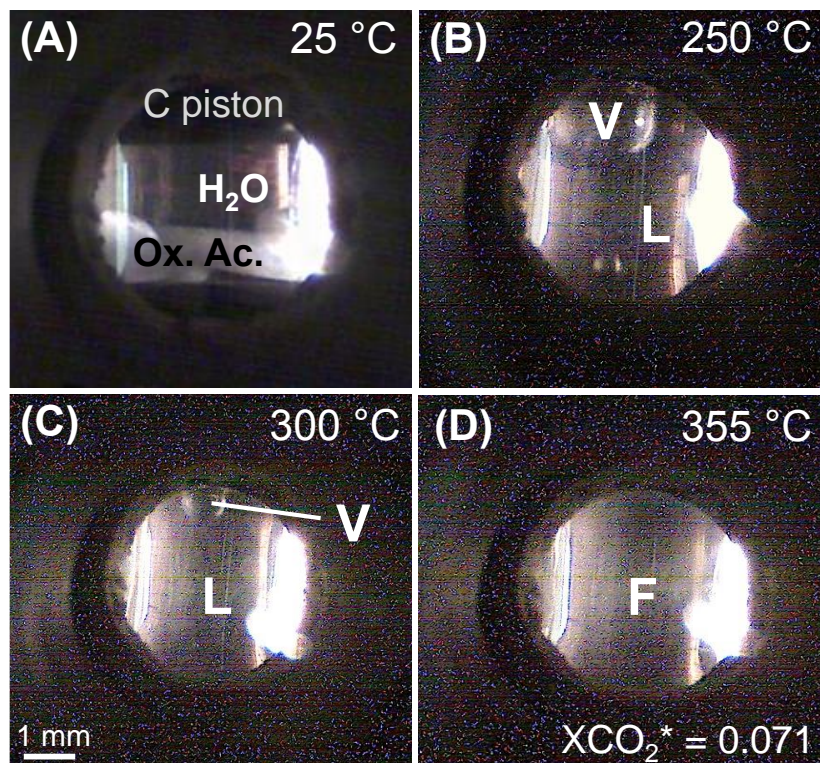
Sapphire windows and cells:



The autoclave : Concept and set-up for *in-situ* spectroscopy

➤ *in-situ* Raman

H₂O – CO₂ (500 bars)



Some perspectives and future developments

The autoclave : Advantages and Limitations

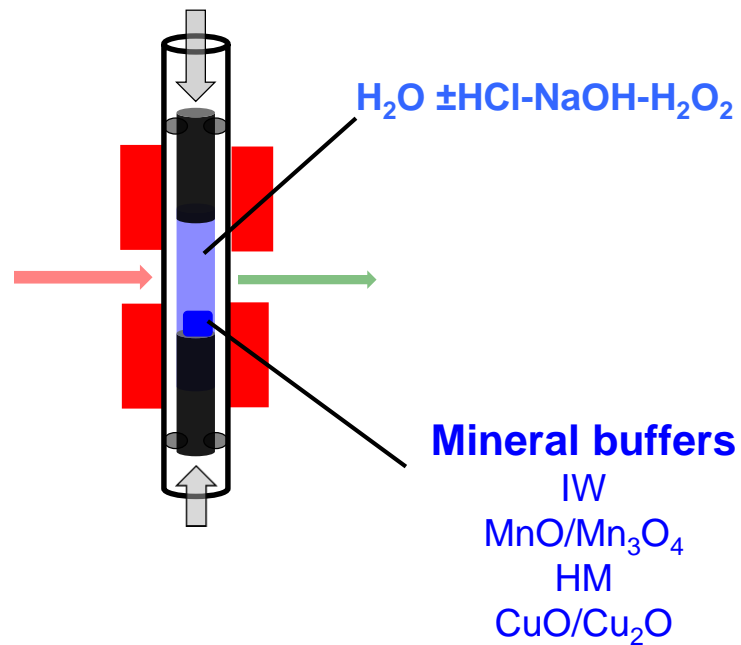
- **Pressure and temperature control** ($\pm 1\text{-}2$ bars, while it is $\pm 0.3\text{-}0.5$ GPa for hydrothermal DAC)

The autoclave : Advantages and Limitations

- **Pressure and temperature control** ($\pm 1\text{-}2$ bars, while it is $\pm 0.3\text{-}0.5$ GPa for hydrothermal DAC)
- **Fluid composition at high P-T** (*Raman : Volatiles speciation - CO_2 , CH_4 , SO_2 , S_3^- , etc...*)

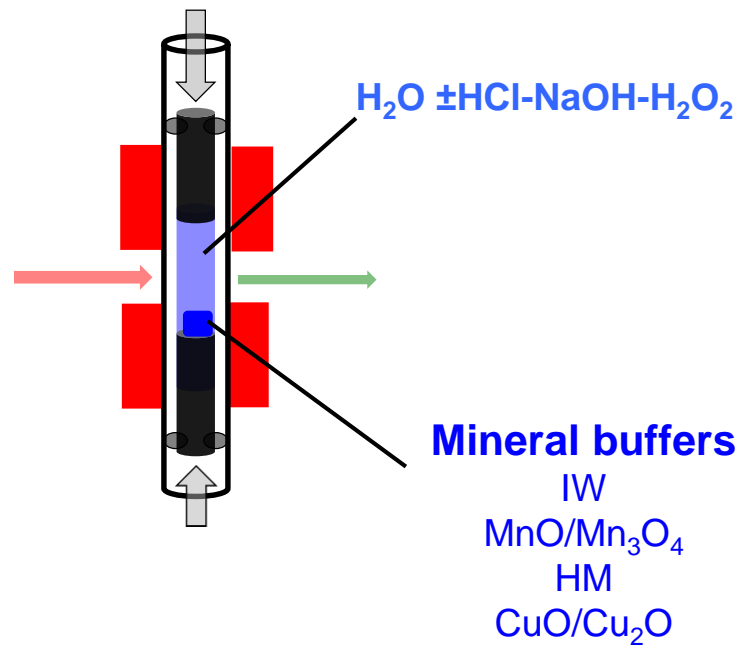
The autoclave : Advantages and Limitations

- **Pressure and temperature control** ($\pm 1\text{-}2$ bars, while it is $\pm 0.3\text{-}0.5$ GPa for hydrothermal DAC)
- **Fluid composition at high P-T** (Raman : Volatiles speciation - CO_2 , CH_4 , SO_2 , S_3^- , etc...)
- **fO₂ control**

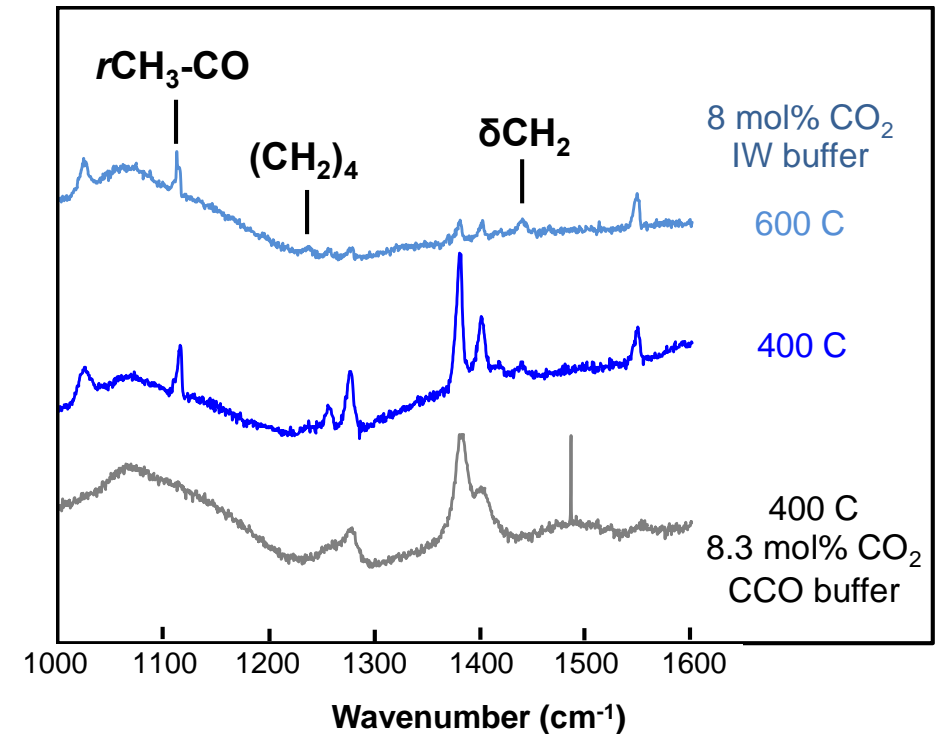


The autoclave : Advantages and Limitations

- Pressure and temperature control ($\pm 1\text{-}2$ bars, while it is $\pm 0.3\text{-}0.5$ GPa for hydrothermal DAC)
- Fluid composition at high P-T (Raman : Volatiles speciation - CO_2 , CH_4 , SO_2 , S_3^- , etc...)
- **fO₂ control**

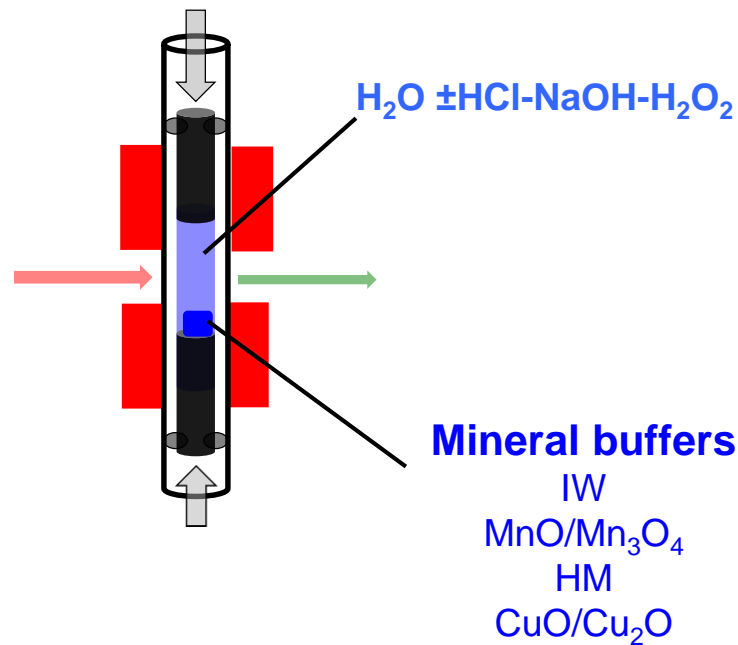


Formation of CH-CO species in H₂O-Ox.Acid mixture at the IW buffer

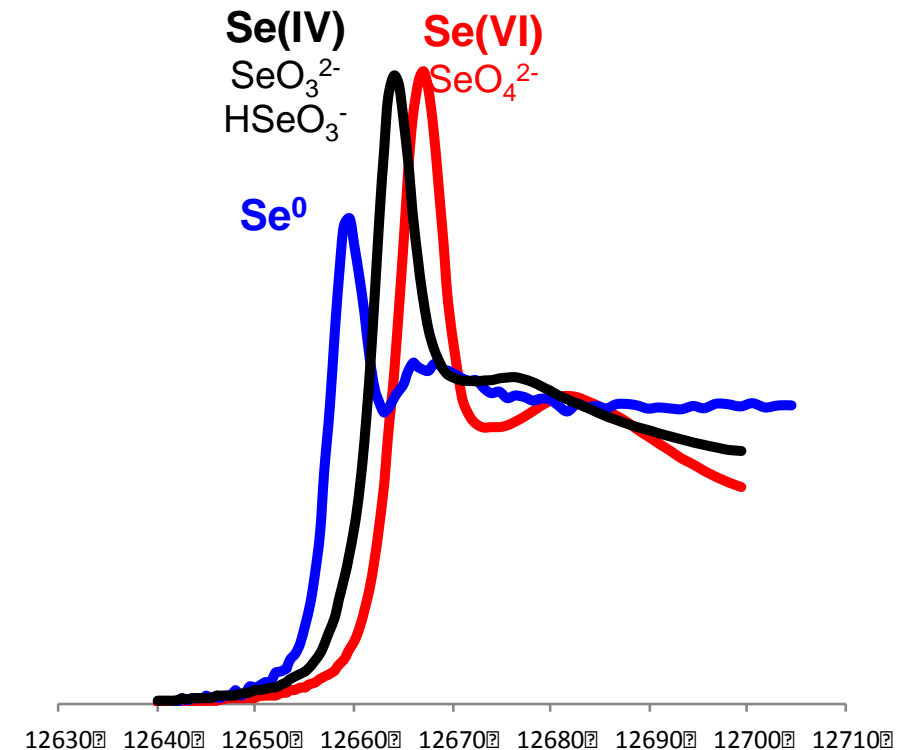


The autoclave : Advantages and Limitations

- Pressure and temperature control ($\pm 1\text{-}2$ bars, while it is $\pm 0.3\text{-}0.5$ GPa for hydrothermal DAC)
- Fluid composition at high P-T (Raman : Volatiles speciation - CO_2 , CH_4 , SO_2 , S_3^- , etc...)
- **fO₂ control**



Se oxidation at $\text{CuO/Cu}_2\text{O}$ buffer with increasing T (100-350 °C)

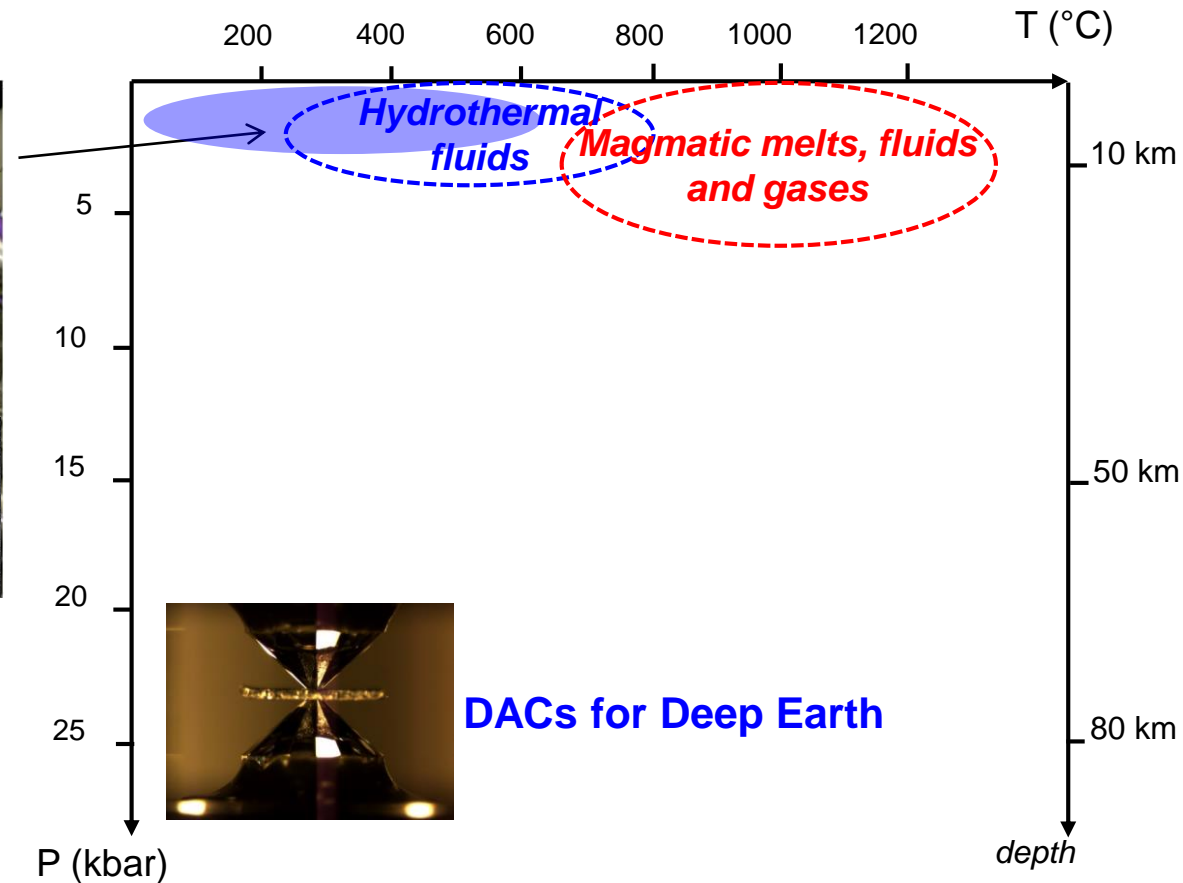


The autoclave : Advantages and Limitations

- **Detection limits**
 - ~ 10-100 ppm for quantification (solubility)
 - ~ 1000 ppm for 'correct' EXAFS

The autoclave : Advantages and Limitations

- **Detection limits** *~ 10-100 ppm for quantification (solubility)*
~ 1000 ppm for 'correct' EXAFS
- **P-T limitations** *only enable moderate P-T*



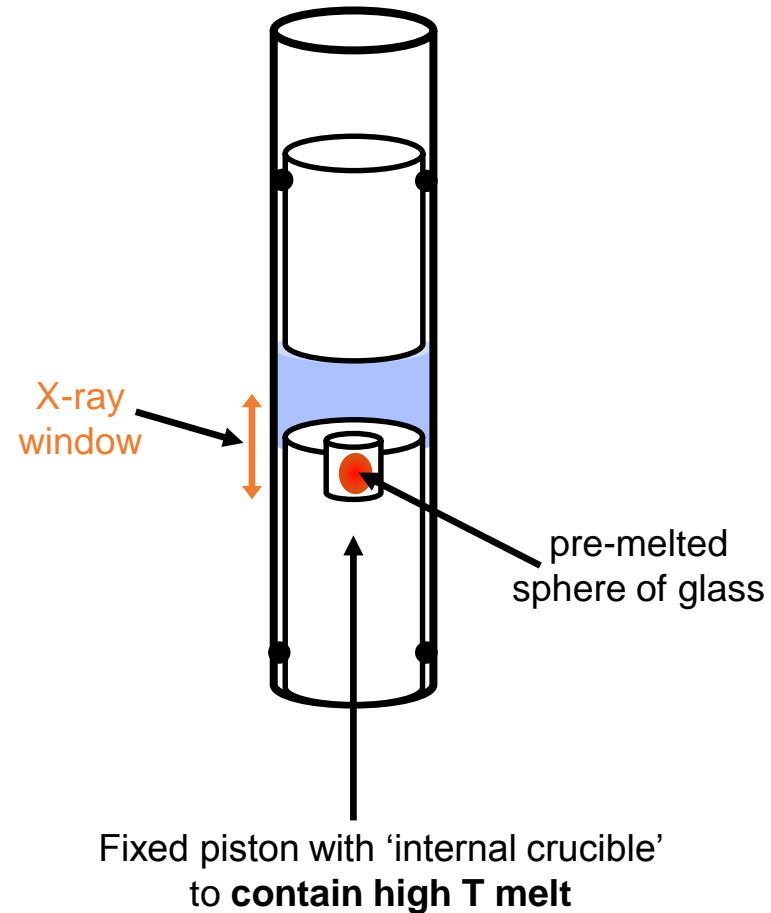
The autoclave : Advantages and Limitations

➤ Detection limits

- ✓ EBS => Improved flux
 - ✓ BM16 => HERFD -XAS
- Ask D. Testemale and E. Bazarkina!*

The autoclave : Advantages and Limitations

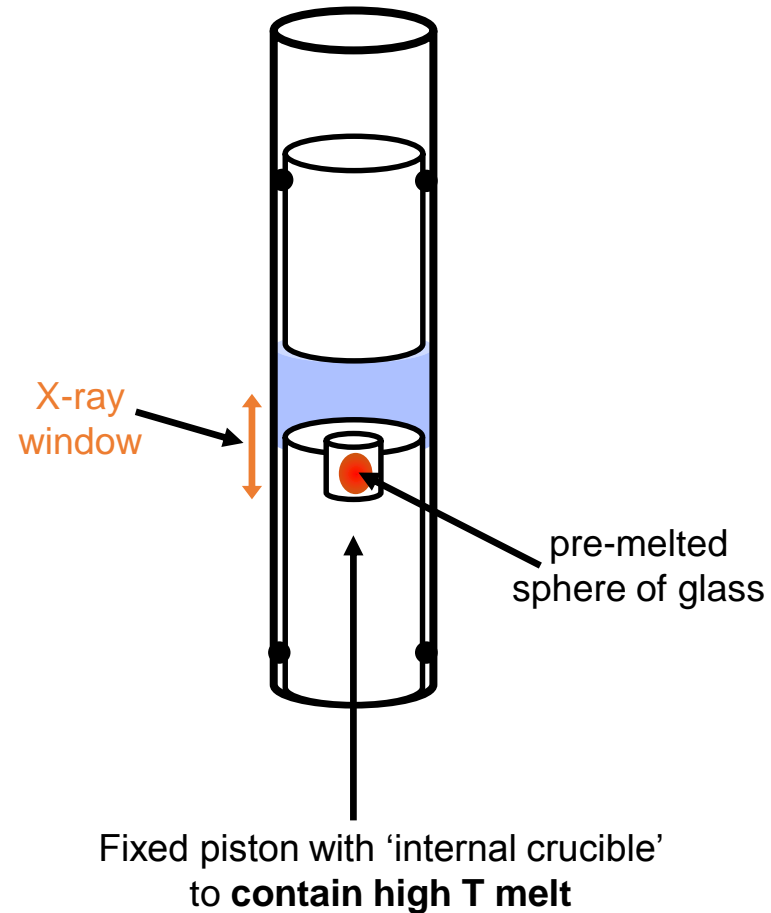
- **P-T limitations** *new Autoclave that can reach 1200 °C !*



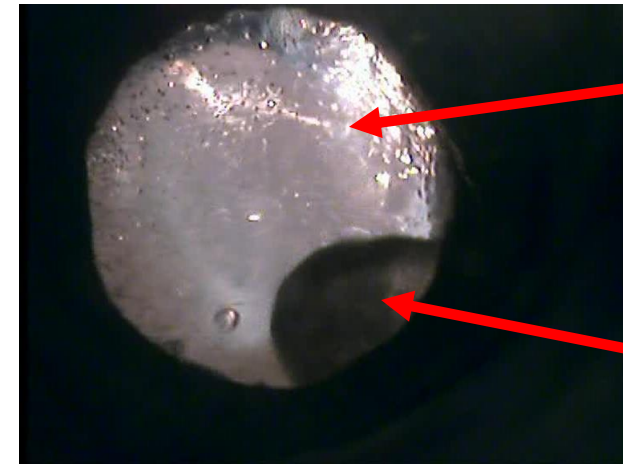
The autoclave : Advantages and Limitations

➤ P-T limitations

new Autoclave that can reach 1200 °C !



800 °C – 1.5 kbar

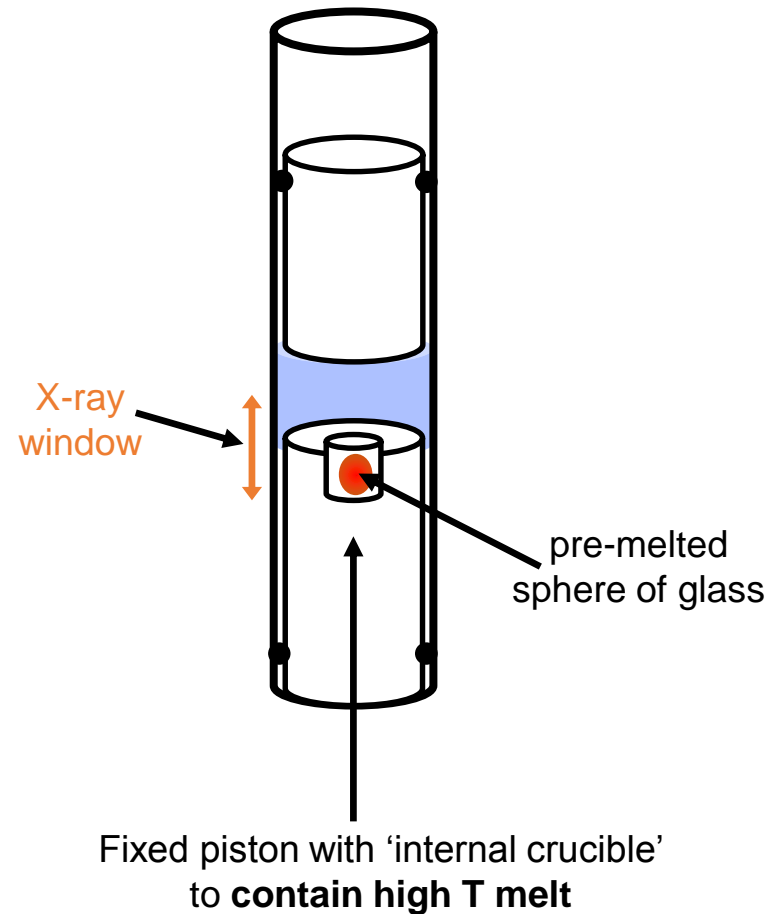


H₂O_{gas}

Haplogranite melt

The autoclave : Advantages and Limitations

➤ **P-T limitations** *new Autoclave that can reach 1200 °C !*



Haplogranite + NaBr sol
650 °C – 1kbar

